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IMIDAZOPYRIDINE-DERIVATIVES AS INDUCIBLE NO-SYNTHASE INHIBITORS

Field of application of the invention

The invention relates to novel imidazopyridine derivatives, which are used in the pharmaceutical industry for the production of pharmaceutical compositions.

Known technical background

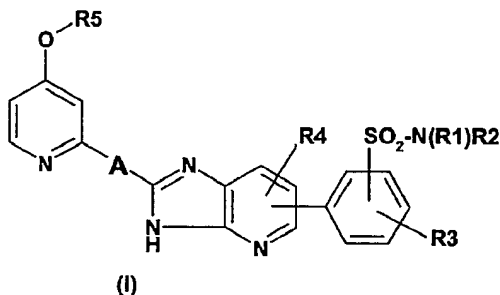
In the German Patent Application DE 2504252 and in the European Patent Application EP 0125756 3H-imidazo[4,5-b]pyridine derivatives with anti-ulcer activity are described.

The International Application WO 0049015 describes pyridine compounds with inhibitory activity on the production of nitric oxide.

Description of the invention

It has now been found that the novel aminosulphonylphenyl-substituted imidazopyridine derivatives, which are described in greater details below, have surprising and particularly advantageous properties.

The invention thus relates, in a first embodiment (embodiment a), to compounds of formula I



in which

R1 is hydrogen or 1-4C-alkyl,

R2 is hydrogen or 1-4C-alkyl, and

R3 is 1-4C-alkyl, trifluoromethyl, or completely or predominantly fluorine-substituted 1-4C-alkoxy;

or in which

R1 is 3-7C-cycloalkyl, phenyl-1-4C-alkyl, hydroxy-2-4C-alkyl, 1-4C-alkoxy-2-4C-alkyl, phenyl, pyridyl, or R11- and/or R12-substituted phenyl, in which

R11 is 1-4C-alkyl, halogen, 1-4C-alkoxy, or mono- or di-1-4C-alkylamino,

R12 is 1-4C-alkyl or halogen,

R2 is hydrogen, hydroxy-2-4C-alkyl, 1-4C-alkoxy-2-4C-alkyl or 1-4C-alkyl, and

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R3 is hydrogen, halogen, 1-4C-alkoxy, 1-4C-alkyl, trifluoromethyl, or completely or predominantly fluorine-substituted 1-4C-alkoxy;

or in which

R1 and R2 together and with inclusion of the nitrogen atom, to which they are bonded, form a heterocyclic ring Het, in which

Het is a fully saturated or partially unsaturated mono- or fused bicyclic ring or ring system made up of

a first constituent being a 3- to 7-membered monocyclic fully saturated non-aromatic heterocyclic ring B,

which heterocyclic ring B comprises one to three heteroatoms independently selected from nitrogen, oxygen and sulfur,

and which heterocyclic ring B is optionally substituted by one or two oxo groups, and, optionally, fused to said first constituent,

a second constituent being a benzene ring,

and which ring Het is optionally substituted by R21 on a ring carbon atom,

and/or which ring Het is optionally substituted by R22 on a further ring carbon atom,

and/or which ring Het is optionally substituted by an ethylenedioxy group,

and/or which ring Het is optionally substituted by R23 on a ring nitrogen atom,

in which

R21 is 1-4C-alkyl, 1-4C-alkoxy or phenylcarbonyl,

R22 is 1-4C-alkyl or 1-4C-alkoxy,

R23 is 1-4C-alkyl, phenyl-1-4C-alkyl, 1-4C-alkylcarbonyl, 1-4C-alkoxy-2-4C-alkyl, mono- or di-1-4C-alkylamino-2-4C-alkyl, phenyl, pyrimidyl, pyridyl, formyl, 3-7C-cycloalkyl, 3-7C-cycloalkylmethyl, or R231- and/or R232-substituted phenyl, in which

R231 is halogen, cyano or 1-4C-alkyl,

R232 is halogen or 1-4C-alkyl, and

R3 is hydrogen, halogen, 1-4C-alkoxy, 1-4C-alkyl, trifluoromethyl, or completely or predominantly fluorine-substituted 1-4C-alkoxy;

and in which

R4 is hydrogen, halogen, 1-4C-alkyl or 1-4C-alkoxy,

R5 is 1-4C-alkyl,

A is 1-4C-alkylene;

the salts, N-oxides and the salts of the N-oxides of these compounds.

The invention relates, in a second embodiment (embodiment b), to compounds of formula I, in which

R1 is hydrogen or 1-4C-alkyl,

R2 is hydrogen or 1-4C-alkyl, and

R3 is 1-4C-alkyl, trifluoromethyl, or completely or predominantly fluorine-substituted 1-4C-alkoxy;

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or in which

R1 is 3-7C-cycloalkyl, phenyl-1-4C-alkyl, hydroxy-2-4C-alkyl, 1-4C-alkoxy-2-4C-alkyl, phenyl, pyridyl, or R11- and/or R12-substituted phenyl, in which

R11 is 1-4C-alkyl, halogen, 1-4C-alkoxy, or mono- or di-1-4C-alkylamino,

R12 is 1-4C-alkyl or halogen,

R2 is hydrogen, hydroxy-2-4C-alkyl, 1-4C-alkoxy-2-4C-alkyl or 1-4C-alkyl, and

R3 is hydrogen, halogen, 1-4C-alkoxy, 1-4C-alkyl, trifluoromethyl, or completely or predominantly fluorine-substituted 1-4C-alkoxy;

or in which

R1 and R2 together and with inclusion of the nitrogen atom, to which they are bonded, form a heterocyclic ring Het, in which

Het is a 3- to 10-membered saturated or partially saturated heterocyclic ring comprising totally 1 to 3 heteroatoms selected from a group consisting of oxygen, sulfur and nitrogen, and optionally substituted by R21 on a ring carbon atom and/or by R22 on a further ring carbon atom and/or by R23 on a ring nitrogen atom, in which

R21 is 1-4C-alkyl, 1-4C-alkoxy or phenylcarbonyl,

R22 is 1-4C-alkyl or 1-4C-alkoxy,

R23 is 1-4C-alkyl, phenyl-1-4C-alkyl, 1-4C-alkylcarbonyl, 1-4C-alkoxy-2-4C-alkyl, mono- or di-1-4C-alkylamino-2-4C-alkyl, phenyl, pyrimidyl, pyridyl, formyl, 3-7C-cycloalkyl, 3-7C-cycloalkylmethyl, or R231- and/or R232-substituted phenyl, in which

R231 is halogen, cyano or 1-4C-alkyl,

R232 is halogen or 1-4C-alkyl, and

R3 is hydrogen, halogen, 1-4C-alkoxy, 1-4C-alkyl, trifluoromethyl, or completely or predominantly fluorine-substituted 1-4C-alkoxy;

and in which

R4 is hydrogen, halogen, 1-4C-alkyl or 1-4C-alkoxy,

R5 is 1-4C-alkyl,

A is 1-4C-alkylene;

the salts, N-oxides and the salts of the N-oxides of these compounds.

1-4C-Alkyl is a straight-chain or branched alkyl radical having 1 to 4 carbon atoms. Examples are the butyl, isobutyl, sec-butyl, tert-butyl, propyl, isopropyl, and, particularly, the ethyl and methyl radicals.

2-4C-Alkyl is a straight-chain or branched alkyl radical having 2 to 4 carbon atoms. Examples are the butyl, isobutyl, sec-butyl, tert-butyl, propyl, isopropyl and, particularly, ethyl radical.

1-4C-Alkylene is a straight chain alkylene radical having 1 to 4 carbon atoms. Examples which may be mentioned in this context are the methylene (-CH₂-), ethylene (-CH₂-CH₂-), trimethylene (-CH₂-CH₂-CH₂-) and the tetramethylene (-CH₂-CH₂-CH₂-CH₂-) radical.

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1-4C-Alkoxy is a radical which, in addition to the oxygen atom, contains a straight-chain or branched alkyl radical having 1 to 4 carbon atoms. Alkoxy radicals having 1 to 4 carbon atoms which may be mentioned in this context are, for example, the butoxy, isobutoxy, sec-butoxy, tert-butoxy, propoxy, isopropoxy, and, particularly, the ethoxy and methoxy radicals.

3-7C-Cycloalkyl stands for cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl and cycloheptyl, of which cyclopropyl, cyclobutyl and cyclopentyl are preferred.

3-7C-Cycloalkylmethyl stands for a methyl radical, which is substituted by one of the abovementioned 3-7C-cycloalkyl radicals. Examples which may be mentioned are the cyclopropylmethyl and the cyclohexylmethyl radicals.

Halogen within the meaning of the present invention is bromine, or preferably chlorine or fluorine.

Completely or predominantly fluorine-substituted 1-4C-alkoxy is, for example, the 2,2,3,3,3-pentafluoropropoxy, the perfluoroethoxy, the 1,2,2-trifluoroethoxy and in particular the 1,1,2,2-tetrafluoroethoxy, the 2,2,2-trifluoroethoxy, the trifluoromethoxy and the difluoromethoxy radical, of which the difluoromethoxy radical is preferred. "Predominantly" in this connection means that more than half of the hydrogen atoms of the 1-4C-alkoxy groups are replaced by fluorine atoms.

1-4C-Alkoxy-2-4C-alkyl stands for one of the abovementioned 2-4C-alkyl radicals which is substituted by one of the abovementioned 1-4C-alkoxy radicals. Examples which may be mentioned are the 2-(methoxy)ethyl ($-\text{CH}_2-\text{CH}_2-\text{O}-\text{CH}_3$), the 3-(methoxy)propyl ($-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{O}-\text{CH}_3$), the 2-(ethoxy)ethyl ($-\text{CH}_2-\text{CH}_2-\text{O}-\text{CH}_2-\text{CH}_3$) and the 2-(isopropoxy)ethyl ($-\text{CH}_2-\text{CH}_2-\text{O}-\text{CH}(\text{CH}_3)_2$) radical.

Hydroxy-2-4C-alkyl stands for one of the abovementioned 2-4C-alkyl radicals which is substituted by an hydroxyl radical. Examples which may be mentioned are the 2-hydroxyethyl and the 3-hydroxypropyl radical.

Mono- or Di-1-4C-alkylamino radicals contain in addition to the nitrogen atom, one or two of the abovementioned 1-4C-alkyl radicals. Preferred are the di-1-4C-alkylamino radicals, especially the dimethylamino, the diethylamino and the diisopropylamino radical.

Mono- or Di-1-4C-alkylamino-2-4C-alkyl stands for one of the abovementioned 2-4C-alkyl radicals which is substituted by one of the abovementioned mono- or di-1-4C-alkylamino radicals. An example which may be mentioned is the 2-(dimethylamino)ethyl radical.

Phenyl-1-4C-alkyl stands for one of the abovementioned 1-4C-alkyl radicals, which is substituted by a phenyl radical. Examples which may be mentioned are the phenethyl and the benzyl radical.

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1-4C-Alkylcarbonyl is a carbonyl group to which one of the abovementioned 1-4C-alkyl radicals is bonded. An example is the acetyl [CH₃-C(O)-] radical.

N-oxide denotes the N-oxide on the pyridine which is substituted by -OR₅.

Het refers in a first aspect (aspect a) to a fully saturated or partially unsaturated mono- or fused bicyclic ring or ring system made up of

a first constituent being a 3- to 7-membered monocyclic fully saturated non-aromatic heterocyclic ring B,

which heterocyclic ring B comprises one to three heteroatoms independently selected from nitrogen, oxygen and sulfur,

and which heterocyclic ring B is optionally substituted by one or two oxo groups, and, optionally, fused to said first constituent,

a second constituent being a benzene ring,

and which ring Het is optionally substituted by R₂₁ on a ring carbon atom,

and/or which ring Het is optionally substituted by R₂₂ on a further ring carbon atom,

and/or which ring Het is optionally substituted by an ethylenedioxy group,

and/or which ring Het is optionally substituted by R₂₃ on a ring nitrogen atom.

Examples for Het according to aspect a may include, but are not limited to, aziridinyl, azetidiny, pyrrolidinyl, piperidinyl, homopiperidinyl, pyrazolidinyl, imidazolidinyl, piperazinyl, homopiperazinyl, morpholinyl or thiomorpholinyl,

and the oxo substituted derivatives of the aforementioned examples such as e.g. 2-oxopyrrolidinyl, 2-oxoimidazolidinyl, 2-oxopiperidinyl, 2,5-dioxopyrrolidinyl, 2,5-dioxoimidazolidinyl, 2,6-dioxopiperidinyl, 2-oxopiperazinyl, or 5-oxo-1,4-diazepanyl, as well as thiomorpholine S-oxide or thiomorpholine S,S-dioxide,

and the benzo-fused derivatives of the aforementioned examples such as e.g. indolinyl, isoindolinyl, 1,2,3,4-tetrahydroquinolinyl or 1,2,3,4-tetrahydroisoquinolinyl.

As used herein, the term "oxo" forms a carbonyl moiety when attached at a carbon atom, a sulfoxide moiety when attached to a sulfur atom and a sulfonyl moiety when two of said terms are attached to a sulfur atom.

Het refers in a second aspect (aspect b), which is an embodiment of aspect a, to a 3- to 10-membered saturated or partially saturated heterocyclic ring radical comprising totally 1 to 3 heteroatoms selected from a group consisting of oxygen, sulfur and nitrogen, and optionally substituted by R₂₁ on a ring carbon atom and/or by R₂₂ on a further ring carbon atom and/or by R₂₃ on a ring nitrogen atom.

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Exemplary Het radicals according to aspect b may be optionally substituted by R21 and/or R22 and/or R23 and may include, without being restricted thereto, azetidin-1-yl, pyrrolidin-1-yl, piperazin-1-yl, imidazolidin-1-yl, thiomorpholin-4-yl, homopiperidin-1-yl, homopiperazin-1-yl, indolin-1-yl, isoindolin-1-yl, 1,2,3,4-tetrahydroquinolin-1-yl, piperidin-1-yl, morpholin-4-yl, 1,2,3,4-tetrahydroisoquinolin-2-yl, 1,4-diazepan-5-one-1-yl, piperazin-3-one-1-yl, or 1,4-dioxo-8-azaspiro[4.5]decan-8-yl.

In this context, as more detailed examples for Het according to aspect b can be mentioned, without being restricted thereto,

piperidin-1-yl, morpholin-4-yl, azetidin-1-yl, pyrrolidin-1-yl, or 1,4-dioxo-8-azaspiro[4.5]decan-8-yl.

Additionally, as more detailed examples for Het according to aspect b can be also mentioned, without being restricted thereto,

piperidin-1-yl substituted by R21,

preferably in which

R21 is 1-4C-alkyl or phenylcarbonyl,

such as, for example, 4-methyl-piperidin-1-yl, or 4-benzoyl-piperidin-1-yl;

1,2,3,4-tetrahydroisoquinolin-2-yl substituted by R21 and/or R22, preferably in which

R21 is 1-4C-alkoxy, and

R22 is 1-4C-alkoxy,

such as, for example, 6,7-dimethoxy-1,2,3,4-tetrahydroisoquinolin-2-yl, or 6,7-diethoxy-1,2,3,4-tetrahydroisoquinolin-2-yl;

piperazin-1-yl substituted by R23 on 4-N,

suitably in which

R23 is 1-4C-alkyl, phenyl-1-4C-alkyl, 1-4C-alkylcarbonyl, 1-4C-alkoxy-2-4C-alkyl, mono- or di-1-4C-alkylamino-2-4C-alkyl, phenyl, pyrimidyl, pyridyl, formyl, 3-7C-cycloalkyl, 3-7C-cycloalkylmethyl, or R231- and/or R232-substituted phenyl, in which

R231 is halogen, cyano or 1-4C-alkyl, and

R232 is halogen or 1-4C-alkyl,

and preferably in which

R23 is 1-4C-alkyl, phenyl-1-4C-alkyl, 1-4C-alkylcarbonyl, 1-4C-alkoxy-2-4C-alkyl, phenyl, or R231- and/or R232-substituted phenyl, in which

R231 is halogen, cyano or 1-4C-alkyl, and

R232 is halogen or 1-4C-alkyl,

such as, for example, 4-N-methyl-piperazin-1-yl, 4-N-ethyl-piperazin-1-yl, 4-N-benzyl-piperazin-1-yl, 4-N-(2-phenethyl)-piperazin-1-yl, 4-N-(2-methoxyethyl)-piperazin-1-yl, 4-N-acetyl-piperazin-1-yl, 4-N-phenyl-piperazin-1-yl, 4-N-(3,5-dichlorophenyl)-piperazin-1-yl, 4-N-(4-cyanophenyl)-piperazin-1-yl, 4-N-(4-methylphenyl)-piperazin-1-yl, 4-N-(2-methylphenyl)-piperazin-1-yl, 4-N-(2,4-dimethylphenyl)-piperazin-1-yl, or 4-N-(2,6-dimethylphenyl)-piperazin-1-yl;

1,4-diazepan-5-one-1-yl substituted by R23 on 4-N,

preferably in which

R23 is 1-4C-alkyl or phenyl-1-4C-alkyl,

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such as, for example, 4-N-methyl-1,4-diazepan-5-one-1-yl, 4-N-ethyl-1,4-diazepan-5-one-1-yl, or 4-N-benzyl-1,4-diazepan-5-one-1-yl; or homopiperazin-1-yl substituted by R23 on 4-N, preferably in which R23 is 1-4C-alkyl, such as, for example, 4-N-methyl-homopiperazin-1-yl.

Suitable salts for compounds of the formula I - depending on substitution - are all acid addition salts or all salts with bases. Particular mention may be made of the pharmacologically tolerable inorganic and organic acids and bases customarily used in pharmacy. Those suitable are, on the one hand, water-insoluble and, particularly, water-soluble acid addition salts with acids such as, for example, hydrochloric acid, hydrobromic acid, phosphoric acid, nitric acid, sulphuric acid, acetic acid, citric acid, D-gluconic acid, benzoic acid, 2-(4-hydroxybenzoyl)benzoic acid, butyric acid, sulphasalicylic acid, maleic acid, lauric acid, malic acid, fumaric acid, succinic acid, oxalic acid, tartaric acid, embonic acid, stearic acid, toluenesulphonic acid, methanesulphonic acid or 3-hydroxy-2-naphthoic acid, the acids being employed in salt preparation - depending on whether a mono- or polybasic acid is concerned and depending on which salt is desired - in an equimolar quantitative ratio or one differing therefrom.

On the other hand, salts with bases are - depending on substitution - also suitable. As examples of salts with bases are mentioned the lithium, sodium, potassium, calcium, aluminium, magnesium, titanium, ammonium, meglumine or guanidinium salts, here, too, the bases being employed in salt preparation in an equimolar quantitative ratio or one differing therefrom.

Pharmacologically intolerable salts, which can be obtained, for example, as process products during the preparation of the compounds according to the invention on an industrial scale, are converted into pharmacologically tolerable salts by processes known to the person skilled in the art.

According to expert's knowledge the compounds of the invention as well as their salts may contain, e.g. when isolated in crystalline form, varying amounts of solvents. Included within the scope of the invention are therefore all solvates and in particular all hydrates of the compounds of formula I as well as all solvates and in particular all hydrates of the salts of the compounds of formula I.

A person skilled in the art knows on the base of his/her expert knowledge that the compounds according to this invention can exist, with regard to the fused imidazo ring, in different tautomeric forms such as e.g. in the 1-H form or, preferably, in the 3-H form, which is shown in formula I. The invention includes all conceivable tautomers in pure form as well as in any mixing ratio. Particularly the present invention includes the pure 1-H- and, preferably, 3-H-tautomers as well as any mixtures thereof.

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Compounds according to embodiment a of this invention worthy to be mentioned are compounds of formula I,

in which

R1 is hydrogen or 1-4C-alkyl,

R2 is hydrogen or 1-4C-alkyl, and

R3 is 1-4C-alkyl, trifluoromethyl, or completely or predominantly fluorine-substituted 1-4C-alkoxy;

or in which

R1 is 3-7C-cycloalkyl, phenyl-1-4C-alkyl, hydroxy-2-4C-alkyl, 1-4C-alkoxy-2-4C-alkyl, phenyl, pyridyl, or R11- and/or R12-substituted phenyl, in which

R11 is 1-4C-alkyl, halogen, 1-4C-alkoxy, or di-1-4C-alkylamino,

R12 is 1-4C-alkyl or halogen,

R2 is hydrogen, hydroxy-2-4C-alkyl, 1-4C-alkoxy-2-4C-alkyl or 1-4C-alkyl, and

R3 is hydrogen, halogen, 1-4C-alkyl, trifluoromethyl, or completely or predominantly fluorine-substituted 1-4C-alkoxy;

or in which

R1 and R2 together and with inclusion of the nitrogen atom, to which they are bonded, form a heterocyclic ring Het, in which

Het is a fully saturated or partially unsaturated mono- or fused bicyclic ring or ring system made up of

a first constituent being a 3- to 7-membered monocyclic fully saturated non-aromatic heterocyclic ring B,

which heterocyclic ring B is piperazine, morpholine, thiomorpholine, homopiperazine, piperidine, pyrrolidine or azetidine,

and which heterocyclic ring B is optionally substituted by one or two oxo groups,

and, optionally, fused to said first constituent,

a second constituent being a benzene ring,

and which ring Het is optionally substituted by R21 on a ring carbon atom,

and/or which ring Het is optionally substituted by R22 on a further ring carbon atom,

and/or which ring Het is optionally substituted by an ethylenedioxy group,

and/or which ring Het is optionally substituted by R23 on a ring nitrogen atom,

in which

R21 is 1-4C-alkyl, 1-4C-alkoxy or phenylcarbonyl,

R22 is 1-4C-alkyl or 1-4C-alkoxy,

R23 is 1-4C-alkyl, phenyl-1-4C-alkyl, 1-4C-alkylcarbonyl, 1-4C-alkoxy-2-4C-alkyl, phenyl, or R231- and/or R232-substituted phenyl, in which

R231 is halogen, cyano or 1-4C-alkyl,

R232 is halogen or 1-4C-alkyl, and

R3 is hydrogen, halogen, 1-4C-alkyl, trifluoromethyl, or completely or predominantly fluorine-substituted 1-4C-alkoxy;

and in which

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R4 is hydrogen, or 1-4C-alkyl,

R5 is methyl,

A is ethylene;

the salts, N-oxides and the salts of the N-oxides of these compounds.

Compounds according to embodiment b of this invention worthy to be mentioned are compounds of formula I, in which

R1 is hydrogen or 1-4C-alkyl,

R2 is hydrogen or 1-4C-alkyl, and

R3 is 1-4C-alkyl, trifluoromethyl, or completely or predominantly fluorine-substituted 1-4C-alkoxy;
or

R1 is 3-7C-cycloalkyl, phenyl-1-4C-alkyl, hydroxy-2-4C-alkyl, phenyl, pyridyl, or R11- and/or R12-substituted phenyl, in which

either

R11 is 1-4C-alkyl, 1-4C-alkoxy, or mono- or di-1-4C-alkylamino, and

R12 is halogen,

or

R11 is halogen, 1-4C-alkoxy, or mono- or di-1-4C-alkylamino, and

R12 is 1-4C-alkyl,

R2 is hydrogen, hydroxy-2-4C-alkyl or 1-4C-alkyl, and

R3 is hydrogen, halogen, 1-4C-alkoxy, 1-4C-alkyl, trifluoromethyl, or completely or predominantly fluorine-substituted 1-4C-alkoxy;

or

R1 and R2 together and with inclusion of the nitrogen atom, to which they are bonded, form a heterocyclic ring Het, in which

Het is optionally substituted by R21 on a ring carbon atom and/or by R22 on a further ring carbon atom and/or by R23 on a ring nitrogen atom and is azetidin-1-yl, pyrrolidin-1-yl, piperazin-1-yl, thiomorpholin-4-yl, homopiperidin-1-yl, homopiperazin-1-yl, indolin-1-yl, isoindolin-1-yl, 1,2,3,4-tetrahydroquinolin-2-yl, piperidin-1-yl, morpholin-4-yl, 1,2,3,4-tetrahydroisoquinolin-1-yl, 1,4-diazepan-5-one-1-yl, or 1,4-dioxo-8-azaspiro[4.5]decan-8-yl, in which

R21 is 1-4C-alkyl, 1-4C-alkoxy or phenylcarbonyl,

R22 is 1-4C-alkoxy,

R23 is 1-4C-alkyl, phenyl-1-4C-alkyl, 1-4C-alkylcarbonyl, 1-4C-alkoxy-2-4C-alkyl, phenyl, or R231- and/or R232-substituted phenyl, in which

R231 is halogen, cyano or 1-4C-alkyl,

R232 is halogen or 1-4C-alkyl, and

R3 is hydrogen, halogen, 1-4C-alkoxy, 1-4C-alkyl, trifluoromethyl, or completely or predominantly fluorine-substituted 1-4C-alkoxy;

R4 is hydrogen,

R5 is methyl,

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A is ethylene;

the salts, N-oxides and the salts of the N-oxides of these compounds.

Compounds according to embodiment a of this invention more worthy to be mentioned are compounds of formula I,

in which

R1 is hydrogen or 1-4C-alkyl,

R2 is hydrogen or 1-4C-alkyl, and

R3 is 1-4C-alkyl, trifluoromethyl, or completely or predominantly fluorine-substituted 1-4C-alkoxy;
or in which

R1 is 3-7C-cycloalkyl, phenyl-1-4C-alkyl, hydroxy-2-4C-alkyl, 1-4C-alkoxy-2-4C-alkyl, phenyl, pyridyl, or R11- and/or R12-substituted phenyl, in which

R11 is 1-4C-alkyl, halogen, 1-4C-alkoxy, or di-1-4C-alkylamino,

R12 is 1-4C-alkyl or halogen,

R2 is hydrogen, hydroxy-2-4C-alkyl, 1-4C-alkoxy-2-4C-alkyl or 1-4C-alkyl, and

R3 is hydrogen, halogen, 1-4C-alkyl, trifluoromethyl, or completely or predominantly fluorine-substituted 1-4C-alkoxy;

or in which

R1 and R2 together and with inclusion of the nitrogen atom, to which they are bonded, form a heterocyclic ring Het, in which

Het is piperidinyl, pyrrolidinyl or azetidiny, or
morpholinyl, thiomorpholinyl, S-oxo-thiomorpholinyl or S,S-dioxo-thiomorpholinyl, or
1,2,3,4-tetrahydroisoquinolinyl or di-(1-4C-alkoxy)-1,2,3,4-tetrahydroisoquinolinyl, or
piperidinyl substituted by either ethylenedioxy or R21, or
4N-(R23)-piperazinyl or 4N-(R23)-homopiperazinyl, or
4N-(H)-1,4-diazepan-5-one-1-yl or 4N-(R23)-1,4-diazepan-5-one-1-yl,

in which

R21 is 1-4C-alkyl, or phenylcarbonyl,

R23 is 1-4C-alkyl, phenyl-1-4C-alkyl, 1-4C-alkylcarbonyl, 1-4C-alkoxy-2-4C-alkyl, phenyl, or R231- and/or R232-substituted phenyl, in which

R231 is halogen, cyano or 1-4C-alkyl,

R232 is halogen or 1-4C-alkyl, and

R3 is hydrogen, halogen, 1-4C-alkyl, trifluoromethyl, or completely or predominantly fluorine-substituted 1-4C-alkoxy;

and in which

R4 is hydrogen, or 1-4C-alkyl,

R5 is methyl,

A is ethylene;

the salts, N-oxides and the salts of the N-oxides of these compounds.

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Compounds according to embodiment b of this invention more worthy to be mentioned are compounds of formula I, in which

R1 is hydrogen or 1-4C-alkyl,

R2 is hydrogen or 1-4C-alkyl, and

R3 is 1-4C-alkyl, trifluoromethyl, or completely or predominantly fluorine-substituted 1-4C-alkoxy;
or

R1 is 3-7C-cycloalkyl, phenyl-1-4C-alkyl, hydroxy-2-4C-alkyl, phenyl, pyridyl, or R11- and/or R12-substituted phenyl, in which

either

R11 is 1-4C-alkyl, 1-4C-alkoxy, or mono- or di-1-4C-alkylamino, and

R12 is halogen,

or

R11 is halogen, 1-4C-alkoxy, or mono- or di-1-4C-alkylamino, and

R12 is 1-4C-alkyl,

R2 is hydrogen, hydroxy-2-4C-alkyl or 1-4C-alkyl, and

R3 is halogen, 1-4C-alkyl, trifluoromethyl, completely or predominantly fluorine-substituted 1-4C-alkoxy, or, particularly, hydrogen;

or

R1 and R2 together and with inclusion of the nitrogen atom, to which they are bonded, form a heterocyclic ring Het, in which

Het is piperidin-1-yl, or piperidin-1-yl substituted by R21, in which

R21 is 1-4C-alkyl or phenylcarbonyl,

or

Het is 1,2,3,4-tetrahydroisoquinolin-2-yl substituted by R21 and R22, in which

R21 is 1-4C-alkoxy,

R22 is 1-4C-alkoxy,

or

Het is piperazin-1-yl substituted by R23 on 4-N, in which

R23 is 1-4C-alkyl, phenyl-1-4C-alkyl, 1-4C-alkylcarbonyl, 1-4C-alkoxy-2-4C-alkyl, phenyl, or R231- and/or R232-substituted phenyl, in which

R231 is halogen, cyano or 1-4C-alkyl,

R232 is halogen or 1-4C-alkyl,

or

Het is 1,4-diazepan-5-one-1-yl, or 1,4-diazepan-5-one-1-yl substituted by R23 on 4-N, in which

R23 is 1-4C-alkyl or phenyl-1-4C-alkyl,

or

Het is homopiperazin-1-yl substituted by R23 on 4-N, in which

R23 is 1-4C-alkyl,

or

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Het is morpholin-4-yl, azetidin-1-yl, pyrrolidin-1-yl, or 1,4-dioxo-8-azaspiro[4.5]decan-8-yl, and
 R3 is hydrogen, halogen, 1-4C-alkyl, trifluoromethyl, or completely or predominantly fluorine-substituted 1-4C-alkoxy;

R4 is hydrogen,

R5 is methyl,

A is ethylene;

the salts, N-oxides and the salts of the N-oxides of these compounds.

Compounds according to embodiment a of this invention in particular worthy to be mentioned are compounds of formula I,

in which

R1 is hydrogen or 1-4C-alkyl,

R2 is hydrogen or 1-4C-alkyl, and

R3 is 1-4C-alkyl, trifluoromethyl, or completely or predominantly fluorine-substituted 1-4C-alkoxy;

or in which

R1 is 3-7C-cycloalkyl, phenyl-1-4C-alkyl, hydroxy-2-4C-alkyl, 1-4C-alkoxy-2-4C-alkyl, phenyl, pyridyl, or R11- and/or R12-substituted phenyl, in which

either

R11 is 1-4C-alkyl, 1-4C-alkoxy, or di-1-4C-alkylamino, and

R12 is halogen,

or

R11 is halogen, 1-4C-alkoxy, or di-1-4C-alkylamino, and

R12 is 1-4C-alkyl,

R2 is hydrogen, hydroxy-2-4C-alkyl, 1-4C-alkoxy-2-4C-alkyl or 1-4C-alkyl, and

R3 is hydrogen;

or in which

R1 and R2 together and with inclusion of the nitrogen atom, to which they are bonded, form a heterocyclic ring Het, in which

Het is piperidinyl, pyrrolidinyl or azetidinyl, or
 morpholinyl, thiomorpholinyl, S-oxo-thiomorpholinyl or S,S-dioxo-thiomorpholinyl, or
 1,2,3,4-tetrahydroisoquinolinyl or di-(1-4C-alkoxy)-1,2,3,4-tetrahydroisoquinolinyl, or
 piperidinyl substituted by either ethylenedioxy or R21, or
 4N-(R23)-piperazinyl or 4N-(1-4C-alkyl)-homopiperazinyl, or
 4N-(H)-1,4-diazepan-5-one-1-yl, 4N-(phenyl-1-4C-alkyl)-1,4-diazepan-5-one-1-yl or 4N-(1-4C-alkyl)-1,4-diazepan-5-one-1-yl,

in which

R21 is 1-4C-alkyl, or phenylcarbonyl,

R23 is 1-4C-alkyl, phenyl-1-4C-alkyl, 1-4C-alkylcarbonyl, 1-4C-alkoxy-2-4C-alkyl, phenyl, or R231- and/or R232-substituted phenyl, in which

R231 is halogen, cyano or 1-4C-alkyl,

R232 is halogen or 1-4C-alkyl, and

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R3 is hydrogen, halogen, 1-4C-alkyl, trifluoromethyl, or completely or predominantly fluorine-substituted 1-4C-alkoxy;

and in which

R4 is hydrogen, or 1-4C-alkyl,

R5 is methyl,

A is ethylene;

the salts, N-oxides and the salts of the N-oxides of these compounds.

Compounds according to embodiment b of this invention in particular worthy to be mentioned are compounds of formula I, in which

R1 is methyl,

R2 is methyl, and

R3 is methyl, trifluoromethyl or trifluoromethoxy;

or

R1 is cyclohexyl, benzyl, 2-hydroxyethyl, phenyl, pyridyl, or R11- and/or R12-substituted phenyl, in which

either

R11 is methyl, methoxy or dimethylamino, and

R12 is chlorine or fluorine,

or

R11 is chlorine, fluorine, methoxy or dimethylamino, and

R12 is methyl,

R2 is hydrogen or methyl,

or R1 and R2 are both 2-hydroxyethyl, and

R3 is hydrogen;

or

R1 and R2 together and with inclusion of the nitrogen atom, to which they are bonded, form a heterocyclic ring Het, in which

Het is piperidin-1-yl, or piperidin-1-yl substituted by R21, in which

R21 is methyl or phenylcarbonyl,

or

Het is 1,2,3,4-tetrahydroisoquinolin-2-yl substituted by R21 and R22, in which

R21 is methoxy,

R22 is methoxy,

or

Het is piperazin-1-yl substituted by R23 on 4-N, in which

R23 is methyl, ethyl, benzyl, phenethyl, acetyl, 2-methoxyethyl, phenyl, or R231- and/or R232-substituted phenyl, in which

R231 is chlorine, cyano or methyl,

R232 is chlorine or methyl,

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or

Het is 1,4-diazepan-5-one-1-yl, or 1,4-diazepan-5-one-1-yl substituted by R23 on 4-N, in which R23 is methyl, ethyl or benzyl,

or

Het is homopiperazin-1-yl substituted by R23 on 4-N, in which R23 is methyl,

or

Het is morpholin-4-yl, azetidin-1-yl, pyrrolidin-1-yl, or 1,4-dioxo-8-azaspiro[4.5]decan-8-yl, and R3 is hydrogen, fluorine, chlorine, methyl, trifluoromethyl or trifluoromethoxy;
R4 is hydrogen,
R5 is methyl,
A is ethylene;

the salts, N-oxides and the salts of the N-oxides of these compounds.

Compounds according to embodiment a of this invention in more particular worthy to be mentioned are compounds of formula I,

in which

R1 is methyl,
R2 is methyl, and
R3 is methyl, trifluoromethyl, or trifluoromethoxy;

or in which

R1 is cyclohexyl, cyclobutyl, cyclopropyl, benzyl, 2-hydroxy-ethyl, 2-methoxy-ethyl, phenyl, pyridyl, or R11- and/or R12-substituted phenyl, in which

either

R11 is methyl, methoxy, or dimethylamino, and
R12 is fluorine,

or

R11 is fluorine, chlorine, methoxy, or dimethylamino, and
R12 is methyl,
R2 is hydrogen, 2-hydroxy-ethyl, 2-methoxy-ethyl, or methyl, and
R3 is hydrogen;

or in which

R1 and R2 together and with inclusion of the nitrogen atom, to which they are bonded, form a heterocyclic ring Het, in which

Het is piperidinyl, pyrrolidinyl or azetidinyl, or
morpholinyl, thiomorpholinyl, S-oxo-thiomorpholinyl or S,S-dioxo-thiomorpholinyl, or
1,2,3,4-tetrahydroisoquinolinyl, di-methoxy-1,2,3,4-tetrahydroisoquinolinyl, or di-ethoxy-
1,2,3,4-tetrahydroisoquinolinyl, or
4,4-ethylenedioxy-piperidinyl or 4-(R21)-piperidinyl, or

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4N-(R23)-piperazinyl or 4N-methyl-homopiperazinyl, or
4N-(H)-1,4-diazepan-5-one-1-yl, 4N-benzyl-1,4-diazepan-5-one-1-yl, 4N-methyl-1,4-diazepan-5-one-1-yl, or 4N-ethyl-1,4-diazepan-5-one-1-yl,

in which

R21 is methyl, or phenylcarbonyl,

R23 is methyl, ethyl, benzyl, phenethyl, acetyl, 2-methoxy-ethyl, phenyl, or R231- and/or R232-substituted phenyl, in which

either

R231 is chlorine, cyano or methyl, and

R232 is chlorine,

or

R231 is chlorine, cyano or methyl, and

R232 is methyl, and

R3 is hydrogen, fluorine, chlorine, methyl, trifluoromethyl, or trifluoromethoxy;

and in which

R4 is hydrogen, or methyl,

R5 is methyl,

A is ethylene;

the salts, N-oxides and the salts of the N-oxides of these compounds.

A special embodiment of the compounds of the present invention include those compounds of formula I in which R5 is methyl.

Another special embodiment of the compounds of the present invention include those compounds of formula I in which A is ethylene.

Another special embodiment of the compounds of the present invention include those compounds of formula I in which R5 is methyl and A is ethylene.

Another special embodiment of the compounds of the present invention include those compounds of formula I in which R4 is hydrogen.

Another special embodiment of the compounds of the present invention include those compounds of formula I in which R4 is methyl.

Another special embodiment of the compounds of the present invention include those compounds of formula I in which R4 is hydrogen, R5 is methyl and A is ethylene.

Another special embodiment of the compounds of the present invention include those compounds of formula I in which R4 is methyl, R5 is methyl and A is ethylene.

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Another special embodiment of the compounds of the present invention include those compounds of formula I in which the aminosulphonylphenyl moiety is bonded to the 6-position of the imidazopyridine ring system.

The substituent R3 and the aminosulphonyl radical of compounds according to this invention can be attached in the ortho, meta or para position with respect to the binding position in which the phenyl ring is bonded to the imidazopyridine ring system, whereby a special embodiment of the compounds of the present invention include those compounds of formula I in which the aminosulphonyl radical is attached in the meta or, particularly, para position.

In this context, another embodiment of the compounds of the present invention include those compounds of formula I in which R3 is attached in the ortho or meta position and the aminosulphonyl radical is attached in the para position with respect to the binding position in which the phenyl ring is bonded to the imidazopyridine ring system.

The substituents R11 and R12 of compounds according to this invention can be attached in the ortho, meta or para position with respect to the binding position in which the phenyl ring is bonded to the nitrogen atom.

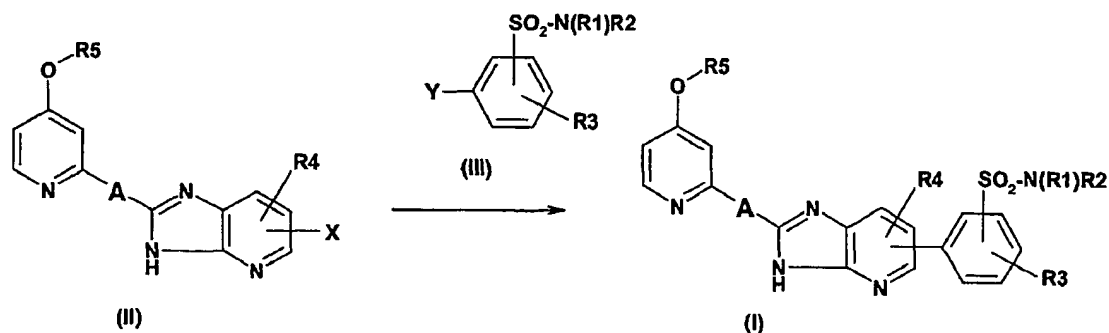
The substituents R231 and R232 of compounds according to this invention can be attached in the ortho, meta or para position with respect to the binding position in which the phenyl ring is bonded to the ring nitrogen atom.

Compounds of formula I can be obtained as described below and shown in the following reaction schemes, or as specified by way of example in the following examples or similarly or analogously thereto.

Thus, as shown in reaction scheme 1 below, a compound of formula II, in which R4, R5 and A have the meanings given above and X is a suitable leaving group, preferably bromine or, particularly, iodine, is reacted with boronic acids or, particularly, boronic acid esters (e.g. pinacol esters) of formula III, in which R1, R2 and R3 have the meanings given above and Y is a boronic acid group or, particularly, a boronic acid ester group, suitably a cyclic boronic acid ester group such as, for example, the boronic acid pinacol ester group, under conditions appropriate for a Suzuki reaction to occur to give compounds of formula I, in which R1, R2, R3, R4, R5 and A have the meanings mentioned above.

Reaction Scheme 1:

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Suitably, the Suzuki reaction is carried out as it is known to the person of ordinary skill in the art and/or in a manner as it is described below and specified by way of example in the following examples or analogously or similarly thereto.

In more detail, the Suzuki reaction mentioned can be carried out in organic solvents alone, for example in toluene, benzene, dimethylformamide or in ethereal (e.g. dimethoxyethane or, in particular, dioxane) or alcohol solvents or in a mixture thereof, or preferably in a mixture comprising an organic solvent (in particular dioxane) and water, with organic (e.g. triethylamine) or preferably inorganic base (e.g. potassium hydroxide, thallium hydroxide, sodium bicarbonate, cesium carbonate, cesium fluoride or, in particular, potassium carbonate) in the presence of a transition metal catalyst, for example, a nickel or, in particular, palladium catalyst (e.g. Pd(OAc)₂, PdCl₂(PPh₃)₂ or, in particular, Pd(PPh₃)₄), and, optionally, lithium chloride. The reaction is carried out at a temperature in the range from 20° to 160°C, usually 60° to 130°C for 10 minutes to 5 days, usually 30 minutes to 24 hours. Advantageously, the solvents used are degassed and the reaction is carried out under protective gas.

The Suzuki reaction is for example described in Tetrahedron Lett. 1998, 39, 4467, J. Org. Chem. 1999, 64, 1372 or Heterocycles 1992, 34, 1395. A general review of Suzuki cross-couplings between boronic acids and aryl halides can be found in Miyaura, N; Suzuki, A. Chem. Rev. 1995, 95, 2457.

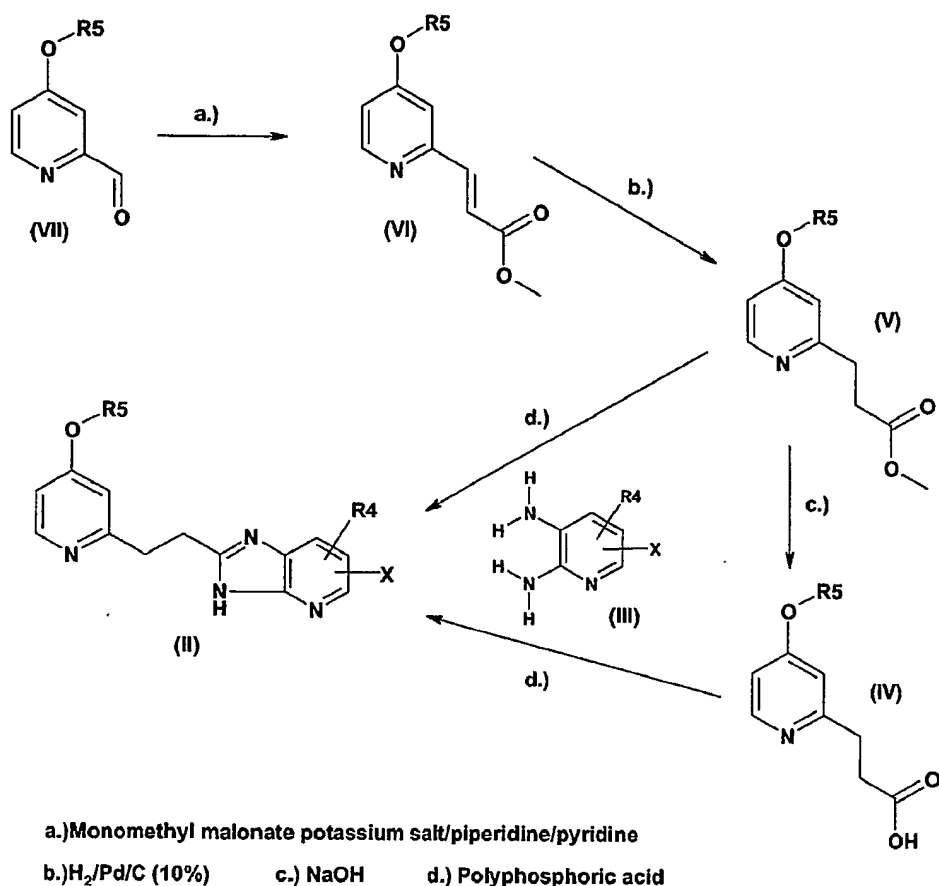
Boronic acids or boronic acid esters (e.g. pinacol esters) of formula III, in which R₁, R₂, R₃ and Y have the meanings given above, are known or can be obtained in an art-known manner or analogously or similarly to known compounds. Boronic acid esters (e.g. pinacol esters) of formula III can be prepared, for example, as described in the following examples starting from phenyl triflates or, particularly, phenyl halides, preferably the bromides or iodides, using e.g. bis-(pinacolato)-diboron in the presence of a transition metal, preferably palladium, catalyst. Optionally the boronic acid esters obtained can be isolated or, preferably, they are generated in situ and used in the subsequent Suzuki reaction without isolation.

Compounds of formula II, in which R₄, R₅, X and A have the meanings given above, are obtained as exemplarily described in the following examples or shown in the following reaction scheme 2 or similarly or analogously thereto.

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In the following reaction scheme 2 the synthesis of compounds of formula II, in which R4, R5 and X have the meanings given above and A is ethylene, is exemplarily described.

Reaction Scheme 2:



The carbon chain in 2-position of the compounds of formula VII is lengthened, for example, by a condensation (with a malonic acid derivative) and a subsequent hydrogenation reaction. Alternatively, the carbon chain can be lengthened using a Wittig reaction followed by a hydrogenation reaction.

The methyl 3-(4-(1-4C)-alkoxypyridin-2-yl)propionate (compound of formula V) or the corresponding acid (compound of formula IV), which can be obtained in an art-known manner, are converted with a 2,3-diaminopyridine derivative (compound of formula III) to give the desired compounds of formula II.

The synthesis of 4-methoxy-pyridin-2-carbaldehyde (compound of formula VII) is described for example in Ashimori et al, Chem Pharm Bull 38, 2446-2458 (1990).

Compounds of formula VII can be also prepared starting from commercially available 4-nitro-2-picoline-N-oxide by exchange of the nitro group by an 1-4C)-alkoxy group. The resulting 4-(1-4C)-

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alkoxy-2-picoline-N-oxide is then via a rearrangement and an oxidation step converted to 4-(1-4C)-alkoxy-pyridin-2-carbaldehyd (compound of formula VII).

The synthesis of 3-(4-methoxypyridin-2-yl)propionic acid (compound of formula IV) is described in the paragraph Starting Materials.

Compounds of formula III, in which R₄ and X have the meanings indicated above, are known or can be prepared in a known manner or analogously or similarly to the preparation of art-known compounds.

Optionally, compounds of formula I can be converted into their salts, or, optionally, salts of the compounds of formula I can be converted into the free compounds. Corresponding processes are known to the person skilled in the art.

The compounds of formula I according to this invention can be converted, optionally, into their N-oxides, for example with the aid of hydrogen peroxide in methanol or with the aid of m-chloroperoxybenzoic acid in dichloromethane. The person skilled in the art is familiar on the basis of his/her expert knowledge with the reaction conditions which are specifically necessary for carrying out the N-oxidation.

It is known to the person skilled in the art that if there are a number of reactive centers on a starting or intermediate compound it may be necessary to block one or more reactive centers temporarily by protective groups in order to allow a reaction to proceed specifically at the desired reaction center. A detailed description for the use of a large number of proven protective groups is found, for example, in T.W. Greene, Protective Groups in Organic Synthesis, John Wiley & Sons, 1999, 3rd Ed, or in P. Kocienski, Protecting Groups, Thieme Medical Publishers, 2000.

The substances according to the invention are isolated and purified in a manner known per se, e.g. by distilling off the solvent in vacuo and recrystallizing the residue obtained from a suitable solvent or subjecting it to one of the customary purification methods, such as column chromatography on a suitable support material.

Salts are obtained by dissolving the free compound in a suitable solvent (for example a ketone like acetone, methylethylketone, or methylisobutylketone, an ether, like diethyl ether, tetrahydrofuran or dioxane, a chlorinated hydrocarbon, such as methylene chloride or chloroform, or a low molecular weight aliphatic alcohol, such as ethanol, isopropanol) which contains the desired acid, or to which the desired acid is then added. The salts are obtained by filtering, reprecipitating, precipitating with a non-solvent for the addition salt or by evaporating the solvent. Salts obtained can be converted by basification into the free compounds which, in turn, can be converted into salts. In this manner, pharmacologically non-tolerable salts can be converted into pharmacologically tolerable salts.

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Suitably, the conversions mentioned in this invention can be carried out analogously or similarly to methods which are familiar per se to the person skilled in the art, for example, in the manner which is described by way of example in the following examples.

The person skilled in the art knows on the basis of his/her knowledge and on the basis of those synthesis routes, which are shown and described within the description of this invention, how to find other possible synthesis routes for compounds according to this invention. All these other possible synthesis routes are also part of this invention.

Having described the invention in detail, the scope of the present invention is not limited only to those described characteristics or embodiments. As will be apparent to persons skilled in the art, modifications, analogies, variations, derivations, homologisations and adaptations to the described invention can be made on the base of art-known knowledge and/or, particularly, on the base of the disclosure (e.g. the explicite, implicate or inherent disclosure) of the present invention without departing from the spirit and scope of this invention as defined by the scope of the appended claims.

The following examples illustrate the invention in greater detail, without restricting it. As well, further compounds according to the present invention, of which the preparation is explicitly not described, can be prepared in an analogous way or in a way which is known by a person skilled in the art using customary preparation methods and process techniques.

The compounds, which are mentioned in the examples as well as their salts are a preferred subject of the invention.

In the examples, m.p. stands for melting point, h for hours, d for days, min for minutes, TLC for thin layer chromatography, R_f for retention factor, MS for mass spectrum, M for molecular ion, other abbreviations have their meanings customary per se for the skilled person.

Examples**Final products****1. 2-[2-(4-Methoxypyridin-2-yl)ethyl]-6-[4-(4-methylpiperazin-1-yl-sulfonyl)-phenyl]-3H-imidazo- [4,5-b]pyridine**

A mixture of 1.12 g of 1-(4-bromo-benzene-sulfonyl)-4-methyl-piperazine, 0.978 g of bis-(pinacolato)-diboron, 0.06 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.077 g of [1,1'-bis(diphenyl-phosphino)ferrocene]palladium-dichloride (complex with CH₂Cl₂), 1.03 g of potassium acetate in 40 ml of degassed dioxane are heated to 90°C under N₂ for 8 hours. To the resulting mixture 15 ml of degassed dioxane, 0.931 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.283 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.678 g of potassium carbonate and 0.208 g of lithium chloride in 15 ml of degassed water are added under N₂. The mixture is heated to reflux under N₂ for 16 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with dichloromethane. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 15-10:1). Concentration of the chromatographically pure fractions and crystallization of the residue from ethylacetate gives 0.545 g of the title compound as a solid of m.p. 193-195°C. The mass spectrum shows the molecular peak MH⁺ at 493.4 Da.

2. 2-[2-(4-Methoxypyridin-2-yl)ethyl]-6-[4-(4-benzylpiperazin-1-yl-sulfonyl)-phenyl]-3H-imidazo- [4,5-b]pyridine

A mixture of 0.593 g of 1-(4-bromo-benzene-sulfonyl)-4-benzyl-piperazine, 0.42 g of bis-(pinacolato)-diboron, 0.025 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.033 g of [1,1'-bis(diphenyl-phosphino)ferrocene]palladium-dichloride (complex with CH₂Cl₂), 0.442 g of potassium acetate in 8 ml of degassed dioxane are heated to 90°C in a sealed tube under N₂ for 16 hours. To the resulting mixture 4 ml of degassed dioxane, 0.371 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.113 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.27 g of potassium carbonate and 0.083 g of lithium chloride in 4 ml of degassed water are added under N₂. The tube is sealed again, the mixture is heated to 110° under N₂ for 17 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with ethylacetate. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 22-20:1 + 1% NEt₃). Concentration of the chromatographically pure fractions and crystallization of the residue from ethylacetate gives 0.128 g of the title compound as a brownish solid of m.p. 160-162°C. The mass spectrum shows the molecular peak MH⁺ at 569.4 Da.

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3. 2-[2-(4-Methoxypyridin-2-yl)ethyl]-6-[4-(4-phenylpiperazin-1-yl-sulfonyl)-phenyl]-3H-imidazo-[4,5-b]pyridine

A mixture of 0.572 g of 1-(4-bromo-benzene-sulfonyl)-4-phenyl-piperazine, 0.42 g of bis-(pinacolato)-diboron, 0.025 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.033 g of [1,1'-bis(diphenylphosphino)ferrocene]palladium-dichloride (complex with CH₂Cl₂), 0.442 g of potassium acetate in 8 ml degassed dioxane are heated to 90°C in a sealed tube under N₂ for 7 hours. To the resulting mixture 4 ml of degassed dioxane, 0.399 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.121 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.29 g of potassium carbonate and 0.089 g of lithium chloride in 4 ml of degassed water are added under N₂. The tube is sealed again, the mixture is heated to 115° under N₂ for 17 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with ethylacetate. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 28-26:1). Concentration of the chromatographically pure fractions and crystallization of the residue from acetonitril/methanol (3:1) gives 0.27 g of the title compound as a brownish solid of m.p. 218-220°C. The mass spectrum shows the molecular peak MH⁺ at 555.4 Da.

4. 2-[2-(4-Methoxypyridin-2-yl)ethyl]-6-[4-[4-(4-cyanophenyl)-piperazin-1-yl-sulfonyl]-phenyl]-3H-imidazo-[4,5-b]pyridine

A mixture of 0.406 g of 1-(4-bromo-benzene-sulfonyl)-4-(4-cyanophenyl)-piperazine, 0.28 g of bis-(pinacolato)-diboron, 0.017 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.022 g of [1,1'-bis(diphenylphosphino)ferrocene]palladium-dichloride (complex with CH₂Cl₂), 0.294 g of potassium acetate in 6 ml of degassed dioxane are heated to 85°C in a sealed tube under N₂ for 18 hours. To the resulting mixture 4 ml of degassed dioxane, 0.228 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.069 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.165 g of potassium carbonate and 0.051 g of lithium chloride in 4 ml of degassed water are added under N₂. The tube is sealed again, the mixture is heated to 110° under N₂ for 5 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with ethylacetate. The combined organic phases are dried over sodium sulfate, concentrated and the residue is crystallized from ethylacetate. Recrystallization from ethylacetate and then from acetonitril gives 0.295 g of the title compound as a brownish solid of m.p. 217-219°C. The mass spectrum shows the molecular peak MH⁺ at 580.5 Da.

5. 2-[2-(4-Methoxypyridin-2-yl)ethyl]-6-[4-(4-p-tolyl-piperazin-1-yl-sulfonyl)-phenyl]-3H-imidazo-[4,5-b]pyridine

A mixture of 0.395 g of 1-(4-bromo-benzene-sulfonyl)-4-(p-tolyl)-piperazine, 0.28 g of bis-(pinacolato)-diboron, 0.017 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.022 g of [1,1'-bis(diphenyl-

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phosphino)ferrocene]palladium-dichloride (complex with CH_2Cl_2), 0.294 g of potassium acetate in 6 ml of degassed dioxane are heated to 85°C in a sealed tube under N_2 for 16 hours. To the resulting mixture 4 ml of degassed dioxane, 0.228 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.069 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.165 g of potassium carbonate and 0.051 g of lithium chloride in 4 ml of degassed water are added under N_2 . The tube is sealed again, the mixture is heated to 115° under N_2 for 7.5 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with ethylacetate. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 25-20:1). Concentration of the chromatographically pure fractions and crystallization of the residue from ethylacetate/diethylether gives 0.128 g of the title compound as a brownish solid of m.p. 150-154°C. The mass spectrum shows the molecular peak MH^+ at 569.5 Da.

6. 6-[4-[4-(2,4-Dimethylphenyl)-piperazin-1-yl-sulfonyl]-phenyl]-2-[2-(4-methoxy-pyridin-2-yl)ethyl]-3H-imidazo[4,5-b]pyridine

A mixture of 0.614 g of 1-(4-bromo-benzene-sulfonyl)-4-(2,4-dimethylphenyl)-piperazine, 0.42 g of bis-(pinacolato)-diboron, 0.025 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.033 g of [1,1'-bis(diphenylphosphino)ferrocene]palladium-dichloride (complex with CH_2Cl_2), 0.442 g of potassium acetate in 8 ml of degassed dioxane are heated to 90°C in a sealed tube under N_2 for 7 hours. To the resulting mixture 10 ml of degassed dioxane, 0.371 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.113 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.27 g of potassium carbonate and 0.083 g of lithium chloride in 10 ml of degassed water are added under N_2 . The tube is sealed again, the mixture is heated to 115° under N_2 for 23 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with ethylacetate. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 25-20:1). Concentration of the chromatographically pure fractions and crystallization of the residue from ethylacetate gives 0.27 g of the title compound as a solid of m.p. 208-209°C. The mass spectrum shows the molecular peak MH^+ at 583.5 Da.

7. 6-[4-[4-(3,5-Dichlorophenyl)-piperazin-1-yl-sulfonyl]-phenyl]-2-[2-(4-methoxy-pyridin-2-yl)ethyl]-3H-imidazo[4,5-b]pyridine

A mixture of 0.675 g of 1-(4-bromo-benzene-sulfonyl)-4-(3,5-dichlorophenyl)-piperazine, 0.42 g of bis-(pinacolato)-diboron, 0.025 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.033 g of [1,1'-bis(diphenylphosphino)ferrocene]palladium-dichloride (complex with CH_2Cl_2), 0.442 g of potassium acetate in 8 ml of degassed dioxane are heated to 85°C in a sealed tube under N_2 for 6 hours. To the resulting mixture 4 ml of degassed dioxane, 0.342 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.104 g of tetrakis(triphenylphosphine)-palladium(0) and

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a solution of 0.25 g of potassium carbonate and 0.076 g of lithium chloride in 4 ml of degassed water are added under N₂. The tube is sealed again, the mixture is heated to 110° under N₂ for 17 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with ethylacetate. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 28-24:1). Concentration of the chromatographically pure fractions and crystallization of the residue from ethylacetate gives 0.047 g of the title compound as a solid of m.p. 220-222°C. The mass spectrum shows the molecular peak MH⁺ at 623.5 Da.

8. 6-[4-[4-(2-Methoxy-ethyl)-piperazin-1-yl-sulfonyl]-phenyl]-2-[2-(4-methoxy-pyridin-2-yl)ethyl]-3H-imidazo[4,5-b]pyridine

A mixture of 0.363 g of 1-(4-bromo-benzene-sulfonyl)-4-(2-methoxy-ethyl)-piperazine, 0.28 g of bis-(pinacolato)-diboron, 0.017 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.022 g of [1,1'-bis(diphenylphosphino)ferrocene]palladium-dichloride (complex with CH₂Cl₂), 0.294 g of potassium acetate in 6 ml degassed dioxane are heated to 85°C in a sealed tube under N₂ for 20 hours. To the resulting mixture 4 ml of degassed dioxane, 0.228 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.69 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.165 g of potassium carbonate and 0.051 g of lithium chloride in 4 ml of degassed water are added under N₂. The tube is sealed again, the mixture is heated to 115° under N₂ for 7 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with dichloromethane. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 25-20:1). Concentration of the chromatographically pure fractions and crystallization of the residue from ethylacetate/diethylether (9:1) gives 0.16 g of the title compound as a solid of m.p. 206-208°C. The mass spectrum shows the molecular peak MH⁺ at 537.4 Da.

9. 6-[4-(4-Acetyl-piperazin-1-yl-sulfonyl)-phenyl]-2-[2-(4-methoxy-pyridin-2-yl)ethyl]-3H-imidazo[4,5-b]pyridine

A mixture of 0.347 g of 1-(4-bromo-benzene-sulfonyl)-4-acetyl-piperazine, 0.28 g of bis-(pinacolato)-diboron, 0.017 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.022 g of [1,1'-bis(diphenylphosphino)ferrocene]palladium-dichloride (complex with CH₂Cl₂), 0.294 g of potassium acetate in 6 ml of degassed dioxane are heated to 85°C in a sealed tube under N₂ for 20 hours. To the resulting mixture 4 ml of degassed dioxane, 0.228 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.069 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.165 g of potassium carbonate and 0.051 g of lithium chloride in 4 ml of degassed water are added under N₂. The tube is sealed again, the mixture is heated to 115° under N₂ for 7 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with ethylacetate. The combined organic phases are dried over sodium sulfate, concentrated and the residue is

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chromatographed on a silica gel column (dichloromethane/methanol 25-20:1). Concentration of the chromatographically pure fractions and crystallization of the residue from ethylacetate/diethylether (9:1) gives 0.175 g of the title compound as a solid of m.p. 138-140°C. The mass spectrum shows the molecular peak MH^+ at 521.4 Da.

10. 2-[2-(4-Methoxypyridin-2-yl)ethyl]-6-[4-(morpholin-4-yl-sulfonyl)-phenyl]-3H-imidazo-[4,5-b]pyridine

A mixture of 0.405 g of 4-(4-bromo-benzene-sulfonyl)-morpholine, 0.42 g of bis-(pinacolato)-diboron, 0.025 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.033 g of [1,1'-bis(diphenylphosphino)ferrocene]palladium-dichloride (complex with CH_2Cl_2), 0.442 g of potassium acetate in 8 ml degassed dioxane are heated to 90°C in a sealed tube under N_2 for 7 hours. To the resulting mixture 4 ml of degassed dioxane, 0.399 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.121 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.29 g of potassium carbonate and 0.089 g of lithium chloride in 4 ml of degassed water are added under N_2 . The tube is sealed again, the mixture is heated to 115° under N_2 for 17 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with ethylacetate. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 30-24:1). Concentration of the chromatographically pure fractions and crystallization of the residue from ethylacetate gives 0.073 g of the title compound as a solid of m.p. 210-212°C. The mass spectrum shows the molecular peak MH^+ at 480.3 Da.

11. 2-[2-(4-Methoxypyridin-2-yl)ethyl]-6-[4-(4-methyl-[1,4]diazepan-1-yl-sulfonyl)-phenyl]-3H-imidazo-[4,5-b]pyridine

A mixture of 0.5 g of 1-(4-bromo-benzene-sulfonyl)-4-methyl-[1,4]diazepane, 0.42 g of bis-(pinacolato)-diboron, 0.025 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.033 g of [1,1'-bis(diphenylphosphino)ferrocene]palladium-dichloride (complex with CH_2Cl_2), 0.442 g of potassium acetate in 6 ml of degassed dioxane are heated to 85°C in a sealed tube under N_2 for 17 hours. To the resulting mixture 5 ml of degassed dioxane, 0.371 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.113 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.27 g of potassium carbonate and 0.083 g of lithium chloride in 5 ml of degassed water are added under N_2 . The tube is sealed again, the mixture is heated to 120° under N_2 for 7 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with dichloromethane. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 22-15:1 + 1% NH_4OH). Concentration of the chromatographically pure fractions, crystallization of the residue from ethylacetate and recrystallization from ethylacetate/acetonitril (4:1) gives 0.275 g of the title compound as a solid of m.p. 160-162°C. The mass spectrum shows the molecular peak MH^+ at 507.3 Da.

12. 2-[2-(4-Methoxypyridin-2-yl)ethyl]-6-[4-(4-methyl-piperidin-1-yl-sulfonyl)-phenyl]-3H-imidazo-[4,5-b]pyridine

A mixture of 0.477 g of 1-(4-bromo-benzene-sulfonyl)-4-methyl-piperidine, 0.42 g of bis-(pinacolato)-diboron, 0.025 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.033 g of [1,1'-bis(diphenylphosphino)ferrocene]palladium-dichloride (complex with CH₂Cl₂), 0.442 g of potassium acetate in 6 ml of degassed dioxane are heated to 85°C in a sealed tube under N₂ for 6 hours. To the resulting mixture 5 ml of degassed dioxane, 0.342 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.104 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.25 g of potassium carbonate and 0.076 g of lithium chloride in 3 ml of degassed water are added under N₂. The tube is sealed again, the mixture is heated to 110° under N₂ for 17 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with dichloromethane. The combined organic phases are dried over sodium sulfate and concentrated. Crystallization of the residue from ethylacetate and recrystallization from acetonitril gives 0.185 g of the title compound as a solid of m.p. 186-188°C. The mass spectrum shows the molecular peak MH⁺ at 492.4 Da.

13. 6-[4-(4-Benzoyl-piperidin-1-yl-sulfonyl)-phenyl]-2-[2-(4-methoxy-pyridin-2-yl)ethyl]-3H-imidazo[4,5-b]pyridine

A mixture of 0.612 g of 1-(4-bromo-benzene-sulfonyl)-4-benzoyl-piperidine, 0.42 g of bis-(pinacolato)-diboron, 0.025 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.033 g of [1,1'-bis(diphenylphosphino)ferrocene]palladium-dichloride (complex with CH₂Cl₂), 0.442 g of potassium acetate in 6 ml of degassed dioxane are heated to 90°C in a sealed tube under N₂ for 6 hours. To the resulting mixture 6 ml of degassed dioxane, 0.371 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.113 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.27 g of potassium carbonate and 0.083 g of lithium chloride in 6 ml of degassed water are added under N₂. The tube is sealed again, the mixture is heated to 115° under N₂ for 19 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with dichloromethane. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 30-25:1). Concentration of the chromatographically pure fractions and crystallization of the residue from ethylacetate gives 0.21 g of the title compound as a solid of m.p. 174-175°C. The mass spectrum shows the molecular peak MH⁺ at 582.3 Da.

14. 6-[4-(1,4-dioxo-8-azaspiro[4.5]decan-8-yl-sulfonyl)-phenyl]-2-[2-(4-methoxy-pyridin-2-yl)ethyl]-3H-imidazo[4,5-b]pyridine

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A mixture of 0.543 g of 8-(4-bromo-benzene-sulfonyl)-1,4-dioxo-8-azaspiro[4.5]decane, 0.42 g of bis-(pinacolato)-diboron, 0.025 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.033 g of [1,1'-bis(diphenylphosphino)ferrocene]palladium-dichloride (complex with CH_2Cl_2), 0.442 g of potassium acetate in 6 ml degassed dioxane are heated to 90°C in a sealed tube under N_2 for 7 hours. To the resulting mixture 8 ml of degassed dioxane, 0.371 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.113 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.27 g of potassium carbonate and 0.083 g of lithium chloride in 8 ml of degassed water are added under N_2 . The tube is sealed again, the mixture is heated to 115° under N_2 for 17 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with dichloromethane. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 30-25:1). Concentration of the chromatographically pure fractions and crystallization of the residue from ethylacetate gives 0.185 g of the title compound as a solid of m.p. 121-122°C. The mass spectrum shows the molecular peak MH^+ at 536.3 Da.

15. 6-[4-(6,7-Dimethoxy-1,2,3,4-tetrahydroisoquinolin-2-yl-sulfonyl)-phenyl]-2-[2-(4-methoxy-pyridin-2-yl)ethyl]-3H-imidazo[4,5-b]pyridine

A mixture of 0.619 g of 2-(4-bromo-benzene-sulfonyl)-6,7-dimethoxy-1,2,3,4-tetrahydro-isoquinoline, 0.42 g of bis-(pinacolato)-diboron, 0.025 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.033 g of [1,1'-bis(diphenylphosphino)ferrocene]palladium-dichloride (complex with CH_2Cl_2), 0.442 g of potassium acetate in 9 ml of degassed dioxane are heated to 90°C in a sealed tube under N_2 for 7 hours. To the resulting mixture 15 ml of degassed dioxane, 0.371 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.113 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.27 g of potassium carbonate and 0.083 g of lithium chloride in 10 ml degassed water are added under N_2 . The tube is sealed again, the mixture is heated to 115° under N_2 for 17 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with dichloromethane. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 30-25:1). Concentration of the chromatographically pure fractions and crystallization of the residue from ethylacetate gives 0.396 g of the title compound as a solid of m.p. 207-208°C. The mass spectrum shows the molecular peak MH^+ at 586.3 Da.

16. 6-[4-(1,4-Diazepan-5-one-1-yl-sulfonyl)-phenyl]-2-[2-(4-methoxy-pyridin-2-yl)ethyl]-3H-imidazo[4,5-b]pyridine

A mixture of 0.50 g of 1-(4-bromo-benzene-sulfonyl)-1,4-diazepan-5-one, 0.42 g of bis-(pinacolato)-diboron, 0.025 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.033 g of [1,1'-bis(diphenylphosphino)ferrocene]palladium-dichloride (complex with CH_2Cl_2), 0.442 g of potassium acetate in 6 ml

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of degassed dioxane are heated to 90°C in a sealed tube under N₂ for 17 hours. To the resulting mixture 5 ml of degassed dioxane, 0.371 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.113 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.27 g of potassium carbonate and 0.083 g of lithium chloride in 5 ml of degassed water are added under N₂. The tube is sealed again, the mixture is heated to 115° under N₂ for 7 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with dichloromethane. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 20-12:1 + 1% NH₄OH). Concentration of the chromatographically pure fractions and crystallization of the residue from acetonitrile gives 0.125 g of the title compound as a solid of m.p. 230-232°C. The mass spectrum shows the molecular peak MH⁺ at 507.3 Da.

17. **N-(2-Hydroxyethyl)-4-{2-[2-(4-methoxypyridin-2-yl)ethyl]-3H-imidazo[4,5-b]pyridin-6-yl}benzenesulfonamid**

A mixture of 0.42 g of N-(2-hydroxyethyl)-4-bromobenzenesulfonamide, 0.42 g of bis-(pinacolato)-diboron, 0.025 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.033 g of [1,1'-bis(diphenylphosphino)-ferrocene]palladium-dichloride (complex with CH₂Cl₂), 0.442 g of potassium acetate in 6 ml of degassed dioxane are heated to 90°C in a sealed tube under N₂ for 7 hours. To the resulting mixture 5 ml of degassed dioxane, 0.371 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.113 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.27 g of potassium carbonate and 0.083 g of lithium chloride in 5 ml of degassed water are added under N₂. The tube is sealed again, the mixture is heated to 115° under N₂ for 17 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with dichloromethane. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 15-8 :1 + 1% NH₄OH). Concentration of the chromatographically pure fractions and crystallisation from ethanol gives 0.20 g of the title compound as a solid of m.p. 218-220°C. The mass spectrum shows the molecular peak MH⁺ at 454.1 Da.

18. **N,N-Bis-(2-hydroxyethyl)-4-{2-[2-(4-methoxypyridin-2-yl)ethyl]-3H-imidazo[4,5-b]pyridin-6-yl}benzenesulfonamid**

A mixture of 0.486 g of N,N-bis-(2-hydroxyethyl)-4-bromobenzenesulfonamide, 0.42 g of bis-(pinacolato)-diboron, 0.025 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.033 g of [1,1'-bis(diphenylphosphino)-ferrocene]palladium-dichloride (complex with CH₂Cl₂), 0.442 g of potassium acetate in 6 ml of degassed dioxane are heated to 90°C in a sealed tube under N₂ for 6.5 hours. To the resulting mixture 5 ml of degassed dioxane, 0.371 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.113 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.27 g of potassium carbonate and 0.083 g of lithium chloride in 5 ml of degassed

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water are added under N₂. The tube is sealed again, the mixture is heated to 115° under N₂ for 17 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with dichloromethane. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 15-8:1 + 1% NH₄OH). Concentration of the chromatographically pure fractions and crystallisation from ethanol gives 0.29 g of the title compound as a solid of m.p. 127-128°C. The mass spectrum shows the molecular peak MH⁺ at 498.3 Da.

19. N-Benzyl-4-{2-[2-(4-methoxypyridin-2-yl)ethyl]-3H-imidazo-[4,5-b]pyridin-6-yl}benzenesulfonamid

A mixture of 0.49 g of N-benzyl-4-bromobenzenesulfonamide, 0.42 g of bis-(pinacolato)-diboron, 0.025 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.033 g of [1,1'-bis(diphenylphosphino)-ferrocene]palladium-dichloride (complex with CH₂Cl₂), 0.442 g of potassium acetate in 6 ml of degassed dioxane are heated to 90°C in a sealed tube under N₂ for 6 hours. To the resulting mixture 5 ml of degassed dioxane, 0.371 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.113 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.27 g of potassium carbonate and 0.083 g of lithium chloride in 5 ml of degassed water are added under N₂. The tube is sealed again, the mixture is heated to 120° under N₂ for 17 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with dichloromethane. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 22-15:1). Concentration of the chromatographically pure fractions and crystallisation from ethylacetate/acetonitril (1:1) gives 0.26 g of the title compound as a solid of m.p. 211-212°C. The mass spectrum shows the molecular peak MH⁺ at 500.3 Da.

20. N-Cyclohexyl-4-{2-[2-(4-methoxypyridin-2-yl)ethyl]-3H-imidazo-[4,5-b]pyridin-6-yl}benzenesulfonamid

A mixture of 0.477 g of N-cyclohexyl-4-bromobenzenesulfonamide, 0.42 g of bis-(pinacolato)-diboron, 0.025 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.033 g of [1,1'-bis(diphenylphosphino)-ferrocene]palladium-dichloride (complex with CH₂Cl₂), 0.442 g of potassium acetate in 6 ml of degassed dioxane are heated to 90°C in a sealed tube under N₂ for 6 hours. To the resulting mixture 5 ml of degassed dioxane, 0.371 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.113 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.27 g of potassium carbonate and 0.083 g of lithium chloride in 5 ml of degassed water are added under N₂. The tube is sealed again, the mixture is heated to 120° under N₂ for 17 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with dichloromethane. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 22-15:1). Concentration of the

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chromatographically pure fractions and crystallisation from ethylacetate/acetonitril (1:1) gives 0.27 g of the title compound as a solid of m.p. 199-200°C. The mass spectrum shows the molecular peak MH^+ at 492.4 Da.

21. 4-[2-[2-(4-Methoxypyridin-2-yl)ethyl]-3H-imidazo[4,5-b]pyridin-6-yl]-N,N-dimethyl-2-trifluoromethoxy-benzenesulfonamide

A mixture of 0.477 g of N,N-dimethyl-4-bromo-2-trifluoromethoxybenzenesulfonamide, 0.42 g of bis-(pinacolato)-diboron, 0.025 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.033 g of [1,1'-bis(diphenylphosphino)-ferrocene]palladium-dichloride (complex with CH_2Cl_2), 0.442 g of potassium acetate in 6 ml of degassed dioxane are heated to 85°C in a sealed tube under N_2 for 19 hours. To the resulting mixture 5 ml of degassed dioxane, 0.371 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.113 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.27 g of potassium carbonate and 0.083 g of lithium chloride in 5 ml of degassed water are added under N_2 . The tube is sealed again, the mixture is heated to 120° under N_2 for 5.5 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with dichloromethane. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 30:26:1). Concentration of the chromatographically pure fractions and crystallisation from ethylacetate gives 0.425 g of the title compound as a solid of m.p. 145-146°C. The mass spectrum shows the molecular peak MH^+ at 522.3 Da.

22. 4-[2-[2-(4-Methoxypyridin-2-yl)ethyl]-3H-imidazo[4,5-b]pyridin-6-yl]-N,N-dimethyl-2-trifluoromethyl-benzenesulfonamide

A mixture of 0.480 g of N,N-dimethyl-4-bromo-2-trifluoromethylbenzenesulfonamide, 0.42 g of bis-(pinacolato)-diboron, 0.025 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.033 g of [1,1'-bis(diphenylphosphino)-ferrocene]palladium-dichloride (complex with CH_2Cl_2), 0.442 g of potassium acetate in 6 ml of degassed dioxane are heated to 90°C in a sealed tube under N_2 for 17 hours. To the resulting mixture 5 ml of degassed dioxane, 0.371 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.113 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.27 g of potassium carbonate and 0.083 g of lithium chloride in 5 ml of degassed water are added under N_2 . The tube is sealed again, the mixture is heated to 120° under N_2 for 7 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with dichloromethane. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 30:25:1). Concentration of the chromatographically pure fractions and crystallisation (twice) from ethylacetate

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gives 0.31 g of the title compound as a solid of m.p. 189-190°C. The mass spectrum shows the molecular peak MH^+ at 506.2 Da.

23. 4-[2-[2-(4-Methoxypyridin-2-yl)ethyl]-3H-imidazo[4,5-b]pyridin-6-yl]-N,N-dimethyl-3-methyl-benzenesulfonamide

A mixture of 0.417 g of N,N-dimethyl-4-bromo-3-methylbenzenesulfonamide, 0.42 g of bis-(pinacolato)-diboron, 0.025 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.033 g of [1,1'-bis(diphenylphosphino)-ferrocene]palladium-dichloride (complex with CH_2Cl_2), 0.442 g of potassium acetate in 6 ml of degassed dioxane are heated to 90°C in a sealed tube under N_2 for 6 hours. To the resulting mixture 5 ml of degassed dioxane, 0.371 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.113 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.27 g of potassium carbonate and 0.083 g of lithium chloride in 5 ml of degassed water are added under N_2 . The tube is sealed again, the mixture is heated to 115° under N_2 for 17 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with dichloromethane. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 30-25:1). Concentration of the chromatographically pure fractions and crystallisation from ethylacetate gives 0.19 g of the title compound as a solid of m.p. 179-180°C. The mass spectrum shows the molecular peak MH^+ at 452.2 Da.

24. 4-[2-[2-(4-Methoxypyridin-2-yl)ethyl]-3H-imidazo[4,5-b]pyridin-6-yl]-N-phenyl-benzenesulfonamide

A mixture of 0.468 g of N-phenyl-4-bromo-benzenesulfonamide, 0.42 g of bis-(pinacolato)-diboron, 0.025 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.033 g of [1,1'-bis(diphenylphosphino)-ferrocene]palladium-dichloride (complex with CH_2Cl_2), 0.442 g of potassium acetate in 6 ml of degassed dioxane are heated to 90°C in a sealed tube under N_2 for 6 hours. To the resulting mixture 5 ml of degassed dioxane, 0.371 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.113 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.27 g of potassium carbonate and 0.083 g of lithium chloride in 5 ml of degassed water are added under N_2 . The tube is sealed again, the mixture is heated to 120° under N_2 for 16 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with dichloromethane. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 28-15:1). Concentration of the chromatographically pure fractions and crystallisation from acetonitril gives 0.125 g of the title

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compound as a solid of m.p. 231-233°C. The mass spectrum shows the molecular peak MH^+ at 507.3 Da.

25. 4-{2-[2-(4-Methoxypyridin-2-yl)ethyl]-3H-imidazo[4,5-b]pyridin-6-yl}-N-p-tolyl-benzenesulfonamide

A mixture of 0.489 g of N-p-tolyl-4-bromo-benzenesulfonamide, 0.42 g of bis-(pinacolato)-diboron, 0.025 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.033 g of [1,1'-bis(diphenylphosphino)-ferrocene]palladium-dichloride (complex with CH_2Cl_2), 0.442 g of potassium acetate in 6 ml of degassed dioxane are heated to 90°C in a sealed tube under N_2 for 17 hours. To the resulting mixture 5 ml of degassed dioxane, 0.371 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.113 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.27 g of potassium carbonate and 0.083 g of lithium chloride in 5 ml of degassed water are added under N_2 . The tube is sealed again, the mixture is heated to 120° under N_2 for 6.5 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with dichloromethane. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 25-15 :1). Concentration of the chromatographically pure fractions and crystallisation from ethylacetate/acetonitril (1:1) gives 0.22 g of the title compound as a solid of m.p. 219-220°C. The mass spectrum shows the molecular peak MH^+ at 500.3 Da.

26. 4-{2-[2-(4-Methoxypyridin-2-yl)ethyl]-3H-imidazo[4,5-b]pyridin-6-yl}-N-(2-methoxyphenyl)-benzenesulfonamide

A mixture of 0.513 g of N-(2-methoxyphenyl)-4-bromo-benzenesulfonamide, 0.42 g of bis-(pinacolato)-diboron, 0.025 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.033 g of [1,1'-bis(diphenylphosphino)-ferrocene]palladium-dichloride (complex with CH_2Cl_2), 0.442 g of potassium acetate in 6 ml of degassed dioxane are heated to 90°C in a sealed tube under N_2 for 17 hours. To the resulting mixture 5 ml of degassed dioxane, 0.371 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.113 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.27 g of potassium carbonate and 0.083 g of lithium chloride in 5 ml of degassed water are added under N_2 . The tube is sealed again, the mixture is heated to 115° under N_2 for 19 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with dichloromethane. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 28-20:1). Concentration of the chromatographically pure fractions and crystallisation from ethylacetate gives 0.14 g of the title compound as a solid of m.p. 215-216°C. The mass spectrum shows the molecular peak MH^+ at 516.3 Da.

27. N-(4-Dimethylamino-phenyl)-4-{2-[2-(4-methoxypyridin-2-yl)ethyl]-3H-imidazo-

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[4,5-b]pyridin-6-yl}benzenesulfonamid

A mixture of 0.400 g of N-(4-dimethylamino-phenyl)-4-bromo-benzenesulfonamide, 0.315 g of bis-(pinacolato)-diboron, 0.019 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.025 g of [1,1'-bis(diphenylphosphino)-ferrocene]palladium-dichloride (complex with CH₂Cl₂), 0.332 g of potassium acetate in 6 ml of degassed dioxane are heated to 90°C in a sealed tube under N₂ for 17 hours. To the resulting mixture 5 ml of degassed dioxane, 0.278 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.085 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.202 g of potassium carbonate and 0.062 g of lithium chloride in 5 ml of degassed water are added under N₂. The tube is sealed again, the mixture is heated to 115° under N₂ for 17 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with dichloromethane. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 25-15:1). Concentration of the chromatographically pure fractions and crystallisation from ethylacetate gives 0.06 g of the title compound as a solid of m.p. 221-222°C. The mass spectrum shows the molecular peak MH⁺ at 529.2 Da.

28. N-(4-Chlorophenyl)-N-methyl-4-[2-[2-(4-methoxypyridin-2-yl)ethyl]-3H-imidazo-
[4,5-b]pyridin-6-yl}benzenesulfonamid

A mixture of 0.541 g of N-(4-chlorophenyl)-N-methyl-4-bromo-benzenesulfonamide, 0.42 g of bis-(pinacolato)-diboron, 0.025 g of 1,1'-bis-(diphenylphosphino)-ferrocene, 0.033 g of [1,1'-bis(diphenylphosphino)-ferrocene]palladium-dichloride (complex with CH₂Cl₂), 0.442 g of potassium acetate in 6 ml of degassed dioxane are heated to 90°C in a sealed tube under N₂ for 17 hours. To the resulting mixture 5 ml of degassed dioxane, 0.371 g of 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1), 0.113 g of tetrakis(triphenylphosphine)-palladium(0) and a solution of 0.27 g of potassium carbonate and 0.083 g of lithium chloride in 5 ml of degassed water are added under N₂. The tube is sealed again, the mixture is heated to 115° under N₂ for 6 hours and, after cooling, addition of water and adjusting the pH to 7, it is extracted three times with dichloromethane. The combined organic phases are dried over sodium sulfate, concentrated and the residue is chromatographed on a silica gel column (dichloromethane/methanol 30-25:1). Concentration of the chromatographically pure fractions and crystallisation from ethylacetate gives 0.17 g of the title compound as a solid of m.p. 191-192°C. The mass spectrum shows the molecular peak MH⁺ at 534.3 Da.

Starting from 2-[2-(4-methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine (starting material A1) and the appropriate boronic acid or boronic acid ester derivatives, which can be prepared in a manner known to the person skilled in the art or analogously or similarly as described in the examples above, such as e.g. in situ from the appropriate bromo-benzenesulfonamide derivatives, the following

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compounds can be obtained according to the procedures as described by way of example in the abovementioned examples or analogously or similarly thereto.

29. 2-[2-(4-Methoxy-pyridin-2-yl)-ethyl]-6-[4-(4-phenethyl-piperazine-1-sulfonyl)-phenyl]-3H-imidazo[4,5-b]pyridine

EF: C₃₂ H₃₄ N₆ O₃ S; MW: calc.: 582.73

MS: fnd.: 583.5 (MH⁺)

30. 6-[4-(4-Ethyl-piperazine-1-sulfonyl)-phenyl]-2-[2-(4-methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridine

EF: C₂₆ H₃₀ N₆ O₃ S; MW: calc.: 506.63

MS: fnd.: 507.4 (MH⁺)

31. 6-[4-[4-(2,6-Dimethyl-phenyl)-piperazine-1-sulfonyl]-phenyl]-2-[2-(4-methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridine

EF: C₃₂ H₃₄ N₆ O₃ S; MW: calc.: 582.73

MS: fnd.: 583.5 (MH⁺)

32. 2-[2-(4-Methoxy-pyridin-2-yl)-ethyl]-6-[4-(4-o-tolyl-piperazine-1-sulfonyl)-phenyl]-3H-imidazo[4,5-b]pyridine

EF: C₃₁ H₃₂ N₆ O₃ S; MW: calc.: 568.70

MS: fnd.: 569.4 (MH⁺)

33. 2-[2-(4-Methoxy-pyridin-2-yl)-ethyl]-6-[3-(4-methyl-piperazine-1-sulfonyl)-phenyl]-3H-imidazo[4,5-b]pyridine

EF: C₂₅ H₂₈ N₆ O₃ S; MW: calc.: 492.60

MS: fnd.: 493.3 (MH⁺)

34. 2-[2-(4-Methoxy-pyridin-2-yl)-ethyl]-6-[4-(piperidine-1-sulfonyl)-phenyl]-3H-imidazo[4,5-b]pyridine

EF: C₂₅ H₂₇ N₅ O₃ S; MW: calc.: 477.59

MS: fnd.: 478.3 (MH⁺)

35. 4-[2-[2-(4-Methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridin-6-yl]-N-phenyl-benzenesulfonamide

EF: C₂₆ H₂₃ N₅ O₃ S; MW: calc.: 485.57

MS: fnd.: 486.2 (MH⁺)

36. 2-[2-(4-Methoxy-pyridin-2-yl)-ethyl]-6-[4-(4-methyl-piperazine-1-sulfonyl)-3-trifluoromethoxy-phenyl]-3H-imidazo[4,5-b]pyridine

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EF: C₂₆ H₂₇ F₃ N₆ O₄ S; MW: calc.: 576.60MS: fnd.: 577.3 (MH⁺)

37. 6,7-Diethoxy-2-(4-{2-[2-(4-methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridin-6-yl}-benzenesulfonyl)-1,2,3,4-tetrahydro-isoquinoline

EF: C₃₃ H₃₅ N₅ O₅ S; MW: calc.: 613.74MS: fnd.: 614.3 (MH⁺)

38. 2-[2-(4-Methoxy-pyridin-2-yl)-ethyl]-6-[4-(4-methyl-piperazine-1-sulfonyl)-3-trifluoromethyl-phenyl]-3H-imidazo[4,5-b]pyridine

EF: C₂₆ H₂₇ F₃ N₆ O₃ S; MW: calc.: 560.60MS: fnd.: 561.2 (MH⁺)

39. 6-[3-Fluoro-4-(4-methyl-piperazine-1-sulfonyl)-phenyl]-2-[2-(4-methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridine

EF: C₂₅ H₂₇ F N₆ O₃ S; MW: calc.: 510.59MS: fnd.: 511.4 (MH⁺)

40. 6-[3-Chloro-4-(4-methyl-piperazine-1-sulfonyl)-phenyl]-2-[2-(4-methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridine

EF: C₂₅ H₂₇ Cl N₆ O₃ S; MW: calc.: 527.05MS: fnd.: 527.3 (MH⁺)

41. 6-[2-Fluoro-4-(4-methyl-piperazine-1-sulfonyl)-phenyl]-2-[2-(4-methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridine

EF: C₂₅ H₂₇ F N₆ O₃ S; MW: calc.: 510.59MS: fnd.: 511.3 (MH⁺)

42. 4-Benzyl-1-(4-{2-[2-(4-methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridin-6-yl}-benzenesulfonyl)-[1,4]diazepan-5-one

EF: C₃₂ H₃₂ N₆ O₄ S; MW: calc.: 596.71MS: fnd.: 597.3 (MH⁺)

43. 4-{2-[2-(4-Methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridin-6-yl}-N-methyl-N-phenyl-benzenesulfonamide

EF: C₂₇ H₂₅ N₅ O₃ S; MW: calc.: 499.60MS: fnd.: 500.20 (MH⁺)

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44. **2-[2-(4-Methoxy-pyridin-2-yl)-ethyl]-6-[2-methyl-4-(4-methyl-piperazine-1-sulfonyl)-phenyl]-3H-imidazo[4,5-b]pyridine**
EF: C₂₆ H₃₀ N₆ O₃ S; MW: calc.: 506.63
MS: fnd.: 507.4 (MH⁺)
45. **1-(4-{2-[2-(4-Methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridin-6-yl}-benzenesulfonyl)-4-methyl-[1,4]diazepan-5-one**
EF: C₂₆ H₂₈ N₆ O₄ S; MW: calc.: 520.61
MS: fnd.: 521.3 (MH⁺)
46. **4-Ethyl-1-(4-{2-[2-(4-methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridin-6-yl}-benzenesulfonyl)-[1,4]diazepan-5-one**
EF: C₂₇ H₃₀ N₆ O₄ S; MW: calc.: 534.64
MS: fnd.: 535.3 (MH⁺)
47. **4-{2-[2-(4-Methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridin-6-yl}-N-o-tolyl-benzenesulfonamide**
EF: C₂₇ H₂₅ N₅ O₃ S; MW: calc.: 499.60
MS: fnd.: 500.2 (MH⁺)
48. **4-{2-[2-(4-Methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridin-6-yl}-N-methyl-N-pyridin-4-yl-benzenesulfonamide**
EF: C₂₆ H₂₄ N₆ O₃ S; MW: calc.: 500.58
MS: fnd.: 501.3 (MH⁺)
49. **4-{2-[2-(4-Methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridin-6-yl}-N-methyl-N-p-tolyl-benzenesulfonamide**
EF: C₂₈ H₂₇ N₅ O₃ S; MW: calc.: 513.62
MS: fnd.: 514.3 (MH⁺)
50. **N-(4-Dimethylamino-phenyl)-4-{2-[2-(4-methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridin-6-yl}-N-methyl-benzenesulfonamide**
EF: C₂₉ H₃₀ N₆ O₃ S; MW: calc.: 542.66
MS: fnd.: 543.2 (MH⁺)
51. **N-(2-Fluoro-4-methyl-phenyl)-4-{2-[2-(4-methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridin-6-yl}-benzenesulfonamide**
EF: C₂₇ H₂₄ F N₅ O₃ S; MW: calc.: 517.59

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MS: fnd.: 518.2 (MH⁺)

52. N-(4-Methoxy-phenyl)-4-{2-[2-(4-methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridin-6-yl}-benzenesulfonamide

EF: C27 H25 N5 O4 S; MW: calc.: 515.60

MS: fnd.: 516.3 (MH⁺)

53. N-(4-Methoxy-phenyl)-4-{2-[2-(4-methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridin-6-yl}-N-methyl-benzenesulfonamide

EF: C28 H27 N5 O4 S; MW: calc.: 529.62

MS: fnd.: 530.2 (MH⁺)

54. 4-{2-[2-(4-Methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridin-6-yl}-N-methyl-N-o-tolyl-benzenesulfonamide

EF: C28 H27 N5 O3 S; MW: calc.: 513.62

MS: fnd.: 514.2 (MH⁺)

55. N-(4-Chloro-phenyl)-4-{2-[2-(4-methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridin-6-yl}-benzenesulfonamide

EF: C26 H22 Cl N5 O3 S; MW: calc.: 520.01

MS: fnd.: 520.1 (MH⁺)

56. 2-[2-(4-Methoxy-pyridin-2-yl)-ethyl]-6-[4-(pyrrolidine-1-sulfonyl)-phenyl]-3H-imidazo[4,5-b]pyridine

EF: C24 H25 N5 O3 S; MW: calc.: 463.56

MS: fnd.: 464.3 (MH⁺)

57. 6-[4-(Azetidine-1-sulfonyl)-phenyl]-2-[2-(4-methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridine

EF: C23 H23 N5 O3 S; MW: calc.: 449.54

MS: fnd.: 450.2 (MH⁺)

58. N,N-Bis-(2-methoxy-ethyl)-4-{2-[2-(4-methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridin-6-yl}-benzenesulfonamide

EF: C26 H31 N5 O5 S; MW: calc.: 525.63

MS: fnd.: 526.4 (MH⁺)

59. N-Cyclobutyl-4-{2-[2-(4-methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridin-6-yl}-benzene-sulfonamide

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EF: C₂₄ H₂₅ N₅ O₃ S; MW: calc.: 463.56MS: fnd.: 464.3 (MH⁺)**60. N-Cyclopropyl-4-{2-[2-(4-methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridin-6-yl}-benzene-sulfonamide**EF: C₂₃ H₂₃ N₅ O₃ S; MW: calc.: 449.54MS: fnd.: 450.2 (MH⁺)**61. 2-[2-(4-Methoxy-pyridin-2-yl)-ethyl]-7-methyl-6-[4-(pyrrolidine-1-sulfonyl)-phenyl]-3H-imidazo[4,5-b]pyridine**EF: C₂₅ H₂₇ N₅ O₃ S; MW: calc.: 477.59MS: fnd.: 478.2 (MH⁺)**62. 2-[2-(4-Methoxy-pyridin-2-yl)-ethyl]-7-methyl-6-[4-(piperidine-1-sulfonyl)-phenyl]-3H-imidazo[4,5-b]pyridine**EF: C₂₆ H₂₉ N₅ O₃ S; MW: calc.: 491.62MS: fnd.: 492.3 (MH⁺)**63. 2-[2-(4-Methoxy-pyridin-2-yl)-ethyl]-7-methyl-6-[4-(morpholine-4-sulfonyl)-phenyl]-3H-imidazo[4,5-b]pyridine**EF: C₂₅ H₂₇ N₅ O₄ S; MW: calc.: 493.59MS: fnd.: 494.3 (MH⁺)**64. 6-[4-(Azetidine-1-sulfonyl)-phenyl]-2-[2-(4-methoxy-pyridin-2-yl)-ethyl]-7-methyl-3H-imidazo-[4,5-b]pyridine**EF: C₂₄ H₂₅ N₅ O₃ S; MW: calc.: 463.56MS: fnd.: 464.3 (MH⁺)**65. 2-[2-(4-Methoxy-pyridin-2-yl)-ethyl]-6-[4-(thiomorpholine-4-sulfonyl)-phenyl]-3H-imidazo[4,5-b]pyridine**EF: C₂₄ H₂₅ N₅ O₃ S₂; MW: calc.: 495.63MS: fnd.: 496.3 (MH⁺)**66. 2-[2-(4-Methoxy-pyridin-2-yl)-ethyl]-6-[4-(1-oxo-1(4)-thiomorpholine-4-sulfonyl)-phenyl]-3H-imidazo[4,5-b]pyridine**EF: C₂₄ H₂₅ N₅ O₄ S₂; MW: calc.: 511.63MS: fnd.: 512.2 (MH⁺)

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67. 6-[4-(1,1-Dioxo-1(6)-thiomorpholine-4-sulfonyl)-phenyl]-2-[2-(4-methoxy-pyridin-2-yl)-ethyl]-3H-imidazo[4,5-b]pyridine

EF: C₂₄ H₂₅ N₅ O₅ S₂; MW: calc.: 527.63

MS: fnd.: 528.2 (MH⁺)

68. 2-(4-{2-[2-(4-Methoxy-pyridin-2-yl)-ethyl]-3-H-imidazo[4,5-b]pyridin-6-yl}-benzenesulfonyl)-1,2,3,4-tetrahydro-isoquinoline

EF: C₂₉ H₂₇ N₅ O₃ S; MW: calc.: 525.63

MS: fnd.:

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Starting materials:**A1. 2-[2-(4-Methoxypyridin-2-yl)ethyl]-6-iodo-3H-imidazo[4,5-b]pyridine**

With stirring, a mixture of 8.06 g of 3-(4-methoxypyridin-2-yl)propionic acid (starting material B1), 9.5 g of 2,3-diamino-5-iodopyridine (Cugola et al., Bioorg.Med. Chem.Lett. 22, 2749-2754 (1996)) and 150 g of polyphosphoric acid (PPA) is heated at 140°C for 22 hours. After cooling, the mixture is poured into about 1000 ml of ice-water and then neutralized (pH 7-8) using 6N aqueous sodium hydroxide solution. The mixture is extracted four times with ethyl acetate and the combined organic phases are evaporated to dryness. The residue is crystallized first from ethyl acetate and then from methanol, giving 9.4 g of the title compound as a light-beige powder of m.p. 207-208°C; the mass spectrum shows the molecular peak MH^+ at 381.2 Da.

B1. 3-(4-Methoxypyridin-2-yl)propionic acid

41.95 g of methyl 3-(4-methoxypyridin-2-yl)propionate (starting material C1) are dissolved in 700 ml of tetrahydrofuran, and 217 ml of 1N sodium hydroxide solution are added. The mixture is stirred at RT until no more starting material is detectable (TLC). The mixture is neutralized using 217 ml of 1N hydrochloric acid solution, evaporated to dryness and dried under high vacuum. The colorless residue is ground and extracted four times with dichloromethane/methanol (9:1). The combined extracts are evaporated to dryness. This gives 33.2 g of the title compound as a colorless powder of m.p. 131-132°C. The mass spectrum shows the molecular peak MH^+ at 182 Da.

C1. Methyl 3-(4-methoxypyridin-2-yl)propionate

43.1 g of methyl 3-(4-methoxypyridin-2-yl)acrylate (starting material D1) in 600 ml of methanol are hydrogenated over 3.0 g of Pd/C (10% strength) until the starting material has disappeared (TLC). The catalyst is filtered off, and the mixture is then concentrated and dried under high vacuum. This gives 41.95 g of the title compound as a light-yellow oil. The mass spectrum shows the molecular peak MH^+ at 196 Da.

D1. Methyl 3-(4-methoxypyridin-2-yl)acrylate

A mixture of 45 g of 4-methoxypyridine-2-carbaldehyde (Ashimori et al., Chem.Pharm.Bull. 38, 2446-2458 (1990)), 75.80 g of pyridine hydrochloride, 102.45 g of monomethyl malonate potassium salt and 4.1 ml of piperidine in 700 ml of pyridine are slowly heated, with stirring, to 120°C. When the evolution of gas starts, the heating source is temporarily removed to stop the reaction from becoming too violent. Once the reaction has subsided, the mixture is stirred at 120°C for a further 2.5 hours, and the pyridine is then distilled off under reduced pressure. The residue is partitioned between ethyl acetate/water and the organic phase is washed with water and dried. The residue obtained after

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concentration is chromatographed on a silica gel column using ethyl acetate/petroleum ether 2:1. This initially gives 43.2 g of the title compound as a yellow oil which crystallizes on standing and then shows a m.p. of 80-82°C. The mass spectrum shows the molecular peak MH^+ at 194 Da.

Commercial applicability

The compounds according to the invention have valuable pharmacological properties which make them commercially utilizable. They are selective inhibitors of the enzyme inducible nitric oxide synthase. Nitric oxide synthases (NO-synthases, NOSs) are enzymes that generate NO and citrulline from the amino acid arginine. In certain pathophysiological situations such as arginine depletion or tetrahydrobiopterin depletion the generation of O_2^- from NO-synthases instead or together with NO has been reported. NO is long known as a signalling molecule in most living organisms including mammals and humans. The most prominent action of NO is its smooth muscle relaxing activity, which is caused on the molecular level by the activation of soluble guanylate cyclase. In the last years a lot of other enzymes have been shown to be regulated by NO or reaction products of NO.

There exist three isoforms of NO-synthases which fall into two classes and differ in their physiologic functions and molecular properties. The first class, known as constitutive NO-synthases, comprises of the endothelial NO-synthase and the neuronal NO-synthase. Both isoenzymes are expressed constitutively in various cell types, but are most prominent in endothelial cells of blood vessel walls (therefore called endothelial NO-synthase, eNOS or NOS-III) and in neuronal cells (therefore called neuronal NO-synthase, nNOS or NOS-I). Activation of these two enzymes is dependent on Ca^{2+} /Calmodulin which is generated by transient increases of the intracellular free Ca^{2+} concentration. Activation of constitutive isoforms leads to transient bursts of nitric oxide resulting in nanomolar cellular or tissue NO concentrations. The endothelial isoform is involved in the physiologic regulation of blood pressure. NO generated by the neuronal isoform seems to have neurotransmitter function and the neuronal isoform is among other regulatory processes involved in memory function (long term potentiation).

In contrast to the constitutive isoforms the activation of inducible NO-synthase (iNOS, NOS-II), the sole member of the second class, is performed by transcriptional activation of the iNOS-promoter. Proinflammatory stimuli lead to transcription of the gene for inducible NO-synthase, which is catalytically active without increases in the intracellular Ca^{2+} -concentration. Due to the long half live of the inducible NO-synthase and the unregulated activity of the enzyme, high micromolar concentrations of NO are generated over longer time periods. These high NO-concentrations alone or in cooperation with other reactive radicals such as O_2^- are cytotoxic. Therefore, in situations of microbial infections, iNOS is involved in cell killing by macrophages and other immune cells during early nonspecific immune responses.

There are a number of pathophysiological situations which among others are characterized by the high expression of inducible NO-synthase and concomitant high NO or O_2^- concentrations. It has been shown that these high NO concentrations alone or in combination with other radical species lead to tissue and organ damage and are causally involved in these pathophysiologies. As inflammation is characterized by the expression of proinflammatory enzymes, including inducible NO-synthase, acute and chronic inflammatory processes are promising diseases for the therapeutic application of

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selective inhibitors of inducible NO-synthase. Other pathophysiologies with high NO-production from inducible NO-synthase are several forms of shock (septic, hemorrhagic and cytokine-induced). It is clear that nonselective NO-synthase inhibitors will lead to cardiovascular and neuronal side effects due to concomitant inhibition of constitutive NO-synthase isoforms.

It has been shown in in-vivo animal models of septic shock that reduction of circulating plasma NO-levels by NO-scavenger or inhibition of inducible NO-synthase restores systemic blood pressure, reduces organ damage and increases survival (deAngelo Exp. Opin. Pharmacother. 19-29, 1999; Redl et al. Shock 8, Suppl. 51, 1997; Strand et al. Crit.Care Med. 26, 1490-1499, 1998). It has also been shown that increased NO production during septic shock contributes to cardiac depression and myocardial dysfunction (Sun et al. J. Mol.Cell Cardiol. 30, 989-997, 1998). Furthermore there are also reports showing reduced infarct size after occlusion of the left anterior coronary artery in the presence of NO-synthase inhibitors (Wang et al. Am. J. Hypertens. 12, 174-182, 1999). Considerable inducible NO-synthase activity is found in human cardiomyopathy and myocarditis, supporting the hypothesis that NO accounts at least in part for the dilatation and impaired contractility in these pathophysiologies (de Belder et al. Br. Heart. J. 4, 426-430, 1995).

In animal models of acute or chronic inflammation, blockade of inducible NO-synthase by isoform-selective or nonselective inhibitors or genetic knock out improves therapeutic outcome. It is reported that experimental arthritis (Connor et al. Eur. J. Pharmacol. 273, 15-24, 1995) and osteoarthritis (Pelletier et al. Arthritis & Rheum. 41, 1275-1286, 1998), experimental inflammations of the gastrointestinal tract (Zingarelli et al. Gut 45, 199-209, 1999), experimental glomerulonephritis (Narita et al. Lab. Invest. 72, 17-24, 1995), experimental diabetes (Corbett et al. PNAS 90, 8992-8995, 1993), LPS-induced experimental lung injury is reduced by inhibition of inducible NO-synthase or in iNOS-knock out mice (Kristof et al. Am. J. Crit. Care. Med. 158, 1883-1889, 1998). A pathophysiological role of inducible NO-synthase derived NO or O_2^- is also discussed in chronic inflammatory diseases such as asthma, bronchitis and COPD.

Furthermore, in models of neurodegenerative diseases of the CNS such as MPTP-induced parkinsonism, amyloid peptide induced Alzheimer's disease (Ishii et al., FASEB J. 14, 1485-1489, 2000), malonate induced Huntington's disease (Connop et al. Neuropharmacol. 35, 459-465, 1996), experimental meningitis (Korytko & Boje Neuropharmacol. 35, 231-237, 1996) and experimental encephalitis (Parkinson et al. J. Mol. Med. 75, 174-186, 1997) a causal participation of NO and inducible NO-synthase has been shown.

Increased iNOS expression has been found in the brains of AIDS victims and it is reasonable to assume a role of iNOS in AIDS related dementia (Bagasra et al. J. Neurovirol. 3 153-167, 1997).

Other studies implicated nitric oxide as a potential mediator of microglia dependent primary demyelination, a hallmark of multiple sclerosis (Parkinson et al. J. Mol. Med. 75, 174-186, 1997).

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An inflammatory reaction with concomitant expression of inducible NO-synthase also takes place during cerebral ischemia and reperfusion (Iadecola et al. Stroke 27, 1373-1380, 1996). Resulting NO together with O_2^- from infiltrating neutrophils is thought to be responsible for cellular and organ damage.

Also, in models of traumatic brain injury (Mesenge et al. J. Neurotrauma 13, 209-214, 1996; Wada et al. Neurosurgery 43, 1427-1436, 1998) NO-synthase inhibitors have been shown to possess protective properties. A regulatory role for inducible NO-synthase has been reported in various tumor cell lines (Tozer & Everett Clin Oncol. 9, 357-264, 1997).

On account of their inducible NO-synthase-inhibiting properties, the compounds according to the invention can be employed in human and veterinary medicine and therapeutics, where an excess of NO or O_2^- due to increases in the activity of inducible NO-synthase is involved. They can be used without limitation for the treatment and prophylaxis of the following diseases:

Acute inflammatory diseases: Septic shock, sepsis, SIRS, hemorrhagic shock, shock states induced by cytokine therapy (IL-2, TNF), organ transplantation and transplant rejection, head trauma, acute lung injury, ARDS, inflammatory skin conditions such as sunburn, inflammatory eye conditions such as uveitis, glaucoma and conjunctivitis.

Chronic inflammatory diseases of peripheral organs and the CNS: gastrointestinal inflammatory diseases such as Crohn's disease, inflammatory bowel disease, ulcerative colitis, lung inflammatory diseases such as asthma and COPD, arthritic disorders such as rheumatoid arthritis, osteoarthritis and gouty arthritis, heart disorders such as cardiomyopathy and myocarditis, arteriosclerosis, neurogenic inflammation, skin diseases such as psoriasis, dermatitis and eczema, diabetes, glomerulonephritis; dementias such as dementias of the Alzheimer's type, vascular dementia, dementia due to a general medical condition, such as AIDS-, Parkinson's disease, Huntington's induced dementias, ALS, multiple sclerosis; necrotizing vasculitides such as polyarteritis nodosa, serum sickness, Wegener's granulomatosis, Kawasaki's syndrome; headaches such as migraine, chronic tension headaches, cluster and vascular headaches, post-traumatic stress disorders; pain disorders such as neuropathic pain; myocardial and cerebral ischemia/reperfusion injury.

The compounds may also be useful in the treatment of cancers that express nitric oxide synthase.

The invention further relates to a method for the treatment of mammals, including humans, which are suffering from one of the abovementioned illnesses. The method is characterized in that a therapeutically active and pharmacologically effective and tolerable amount of one or more of the compounds according to the invention is administered to the ill mammal.

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The invention further relates to the compounds according to the invention for use in the treatment and/or prophylaxis of illnesses, especially the illnesses mentioned.

The invention also relates to the use of the compounds according to the invention for the production of pharmaceutical compositions which are employed for the treatment and/or prophylaxis of the illnesses mentioned.

The invention also relates to the use of the compounds according to the invention for the production of pharmaceutical compositions having an iNOS inhibitory activity.

The invention furthermore relates to pharmaceutical compositions for the treatment and/or prophylaxis of the illnesses mentioned, which contain one or more of the compounds according to the invention.

The invention moreover relates to pharmaceutical compositions according to this invention having an iNOS inhibitory activity.

The pharmaceutical compositions are prepared by processes which are known per se and familiar to the person skilled in the art. As pharmaceutical compositions, the compounds according to the invention (= active compounds) are either employed as such, or preferably in combination with suitable pharmaceutical auxiliaries and/or excipients, e.g. in the form of tablets, coated tablets, capsules, caplets, suppositories, patches (e.g. as TTS), emulsions, suspensions, gels or solutions, the active compound content advantageously being between 0.1 and 95% and where, by the appropriate choice of the auxiliaries and/or excipients, a pharmaceutical administration form (e.g. a delayed release form or an enteric form) exactly suited to the active compound and/or to the desired onset of action can be achieved.

The person skilled in the art is familiar with auxiliaries or excipients which are suitable for the desired pharmaceutical formulations on account of his/her expert knowledge. In addition to solvents, gel formers, ointment bases and other active compound excipients, for example antioxidants, dispersants, emulsifiers, preservatives, solubilizers, colorants, complexing agents or permeation promoters, can be used.

The administration of the pharmaceutical compositions according to the invention may be performed in any of the generally accepted modes of administration available in the art. Illustrative examples of suitable modes of administration include intravenous, oral, nasal, parenteral, topical, transdermal and rectal delivery. Oral and intravenous delivery are preferred.

For the treatment of disorders of the respiratory tract, the compounds according to the invention are preferably also administered by inhalation in the form of an aerosol; the aerosol particles of solid, liquid or mixed composition preferably having a diameter of 0.5 to 10 μm , advantageously of 2 to 6 μm .

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Aerosol generation can be carried out, for example, by pressure-driven jet atomizers or ultrasonic atomizers, but advantageously by propellant-driven metered aerosols or propellant-free administration of micronized active compounds from inhalation capsules.

Depending on the inhaler system used, in addition to the active compounds the administration forms additionally contain the required excipients, such as, for example, propellants (e.g. Frigen in the case of metered aerosols), surface-active substances, emulsifiers, stabilizers, preservatives, flavorings, fillers (e.g. lactose in the case of powder inhalers) or, if appropriate, further active compounds.

For the purposes of inhalation, a large number of apparatuses are available with which aerosols of optimum particle size can be generated and administered, using an inhalation technique which is as right as possible for the patient. In addition to the use of adaptors (spacers, expanders) and pear-shaped containers (e.g. Nebulator®, Volumatic®), and automatic devices emitting a puffer spray (Autohaler®), for metered aerosols, in particular in the case of powder inhalers, a number of technical solutions are available (e.g. Diskhaler®, Rotadisk®, Turbohaler® or the inhaler described in European Patent Application EP 0 505 321), using which an optimal administration of active compound can be achieved.

For the treatment of dermatoses, the compounds according to the invention are in particular administered in the form of those pharmaceutical compositions which are suitable for topical application. For the production of the pharmaceutical compositions, the compounds according to the invention (= active compounds) are preferably mixed with suitable pharmaceutical auxiliaries and further processed to give suitable pharmaceutical formulations. Suitable pharmaceutical formulations are, for example, powders, emulsions, suspensions, sprays, oils, ointments, fatty ointments, creams, pastes, gels or solutions.

The pharmaceutical compositions according to the invention are prepared by processes known per se. The dosage of the active compounds is carried out in the order of magnitude customary for iNOS inhibitors. Topical application forms (such as ointments) for the treatment of dermatoses thus contain the active compounds in a concentration of, for example, 0.1-99%. The dose for administration by inhalation is customarily between 0.1 and 10 mg per day. The customary dose in the case of systemic therapy (p.o.) is between 0.3 and 30 mg/kg per day, (i. v.) is between 0.3 and 30 mg/kg/h.

Biological investigations

Measurement of inducible NO-synthase activity

The assay is performed in 96-well microtiter F-plates (Greiner, Frickenhausen, FRG) in a total volume of 100 μ l in the presence of 100 nM calmodulin, 226 μ M CaCl_2 , 477 μ M MgCl_2 , 5 μ M flavin-adenine-dinucleotide (FAD), 5 μ M flavin mononucleotide (FMN), 0.1 mM NADPH, 7 mM glutathione, 10 μ M BH4 and 100 mM HEPES pH 7.2. Arginine concentrations are 0.1 μ M for enzyme inhibition experiments. 150000 dpm of [^3H]arginine are added to the assay mixture. Enzyme reaction is started by the addition of 4 μ g of a crude cytosolic fraction containing human inducible NO-synthase and the reaction mixture is incubated for 45 to 60 min at 37°C. Enzyme reaction is stopped by adding 10 μ l of 2M MES-buffer pH 5.0. 50 μ l of the incubation mixture are transferred into a MADP N65 filtration microtiter plate (Millipore, Eschborn, FRG) containing already 50 μ l of AG-50W-X8 cation exchange resin (Biorad, München, FRG). The resin in the Na loaded form is pre-equilibrated in water and 70 μ l (corresponding to 50 μ l dry beads) are pipetted under heavy stirring with a 8 channel pipette into the filtration plate. After pipetting 50 μ l of the enzyme reaction mixture onto the filtration plates, the plates are placed on a filtration manifold (Porvair, Shepperton, UK) and the flow through is collected in Pico scintillation plates (Packard, Meriden, CT). The resin in the filtration plates is washed with 75 μ l of water (1x50 μ l and 1x 25 μ l) which is also collected in the same plate as the sample. The total flow through of 125 μ l is mixed with 175 μ l of Microscint-40 scintillation cocktail (Packard) and the scintillation plate is sealed with TopSeal P-foil (Packard). Scintillation plates are counted in a scintillation counter.

For the measurement of inducible NO-synthase-inhibiting potencies of compounds increasing concentrations of inhibitors were included into the incubation mixture. IC_{50} -values were calculated from the percent inhibition at given concentrations by nonlinear least square fitting.

The inhibitory values determined for the compounds according to the invention follow from the following table A, in which the compound numbers correspond to the example numbers.

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Table A**Inhibition of iNOS activity [measured as -logIC₅₀ (mol/l)]**

compound	-logIC ₅₀
1	7.71
2	7.45
3	7.84
4	7.52
5	7.58
6	6.90
7	6.51
8	7.37
9	7.53
10	7.64
11	7.89
12	7.60
13	7.48
14	7.58
15	7.46
16	7.62
17	7.37
18	7.04
19	7.73
20	7.30
21	6.95
22	7.0
23	7.17
24	7.71

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Table A (continuation)

25	7.36
26	7.46
27	7.34
28	7.41
29 to 68	The inhibitory values of these mentioned Examples lie in the range from 6.43 to 7.85